Tetrahedron 66 (2010) 8407-8419

Contents lists available at ScienceDirect

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

Total synthesis of xanthanolides

Kazumasa Matsuo^a, Keiko Ohtsuki^a, Takashi Yoshikawa^a, Kozo Shishido^b, Kaori Yokotani-Tomita^c, Mitsuru Shindo^{d,*}

^a Interdisciplinary Graduate School of Engineering Sciences, Kyushu University, Kasuga-koen, Kasuga 816-8580, Japan
^b Graduate School of Pharmaceutical Sciences, The University of Tokushima, Shomachi, Tokushima 770-8505, Japan
^c Graduate School of Life and Environmental Sciences, University of Tsukuba, Tsukuba, Ibaraki 305-8572, Japan

ABSTRACT

^d Institute for Materials Chemistry and Engineering, Kyushu University, Kasuga-koen, Kasuga 816-8580, Japan

A R T I C L E I N F O

Article history: Received 28 June 2010 Received in revised form 26 August 2010 Accepted 26 August 2010 Available online 18 September 2010

Keywords: Xanthanolides Natural products Sesquiterpene lactones Total synthesis

1. Introduction

The xanthanolide sesquiterpene lactones are a class of natural products isolated from the plants of the genus Xanthium (family Composite). They have a bicyclic 5,7-fused ring system, and are known to exhibit allelopathic,¹ antitumor,² antimalarial,³ and antimicrobial activity.⁴ Accordingly, their intriguing biological profile has attracted the interest of medicinal chemists and biochemists as well as organic chemists. An efficient synthesis of these sesquiterpenes therefore would be extremely useful in order to identify biological targets and develop related drugs. Xanthatin (3), isolated from Xanthium strumarium⁵ and other Xanthium families, has shown potent antibacterial activity against methicillin-sensitive and methicillin-resistant Staphylococcus aureus.⁶ 8-epi-Xanthatin (1), which is found in the extracts of the aerial parts of various species in the genus Xanthium,⁷ has also been reported to exhibit antitumor activity^{2a} and in vitro inhibitory activity on farnesyltransferase,^{2d} insect development,^{7b} and auxin-induced growth of sunflower hypocotyls.^{1b,c} In spite of these attractive bioactivities, only a few synthetic studies of the xanthanolides and their related natural products have been reported (Fig. 1).^{8,9} Recently, we have achieved an efficient total synthesis of a dinorxanthanolide, (+)-sundiversifolide (**4**),^{9b} which was isolated from the exudate of germinating sunflowers (Helianthus annuus L.) as an allelopathic compound,^{1a} via a novel construction of the seven-membered

carbocycle by an intramolecular acylation and a one-pot Wittiglactonization as the key steps. This synthesis afforded useful intermediates having a bicyclic 5,7-fused ring system for the synthesis of the xanthanolides. Herein, we report the total synthesis of (+)-8-*epi*-xanthatin (1), (-)-dihydroxanthatin (2), and (-)-xanthatin (3) starting from the common key intermediate 5 as well as full details of the total synthesis of (+)-sundiversifolide.

© 2010 Elsevier Ltd. All rights reserved.

The total synthesis and determination of the absolute configuration of (+)- and (-)-sundiversifolide have

been achieved via intramolecular acylation and Wittig-lactonization as the key steps. The xanthanolide

sesquiterpene lactones, 8-epi-xanthatin (1), dihydroxanthatin (2), and xanthatin (3) were also prepared,

starting from a common intermediate derived from the synthesis of sundiversifolide.

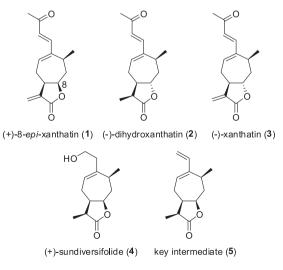


Fig. 1. Xanthanolide natural products.





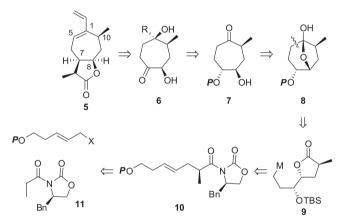
^{*} Corresponding author. Fax: +81 92 583 7875; e-mail address: shindo@ cm.kyushu-u.ac.jp (M. Shindo).

^{0040-4020/\$ –} see front matter @ 2010 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2010.08.061

2. Results and discussion

2.1. Synthesis of sundiversifolide and the key intermediate for the xanthanolides

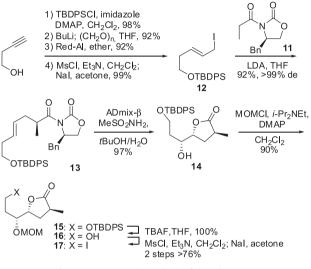
Our synthetic strategy for the key intermediate **5** is illustrated in Scheme 1. The γ -lactone moiety would be formed by acylation-olefination of the hydroxyketone 6. followed by hydrogenation. The vinyl side chain on C-1 would then be introduced by alkylation of the ketone 7. The seven-membered ring construction is much more challenging than the six-membered ring, due to the entropically unfavorable ring size and/or non-bonded interactions in the transition state. To overcome these issues, we envisioned an intramolecular acylation of a carbanion on the γ lactone 9 leading to the 8-oxabicyclo[3.2.1]octane skeleton 8 via a six-membered ring formation. Since the fused oxabicycle 8 is a hemiacetal, the hydroxycycloheptanone 7 would be easily provided. The relevant strategy has been reported by Molander and co-workers utilizing SmI₂.¹⁰ The asymmetric centers on **9** would be obtained by a diastereoselective alkylation using the chiral oxazolidinone 11, followed by stereocontrolled dihydroxylation of the resulting 10.



Scheme 1. Retro synthesis of the key intermediate for the xanthanolides.

The alkylating reagent **12** was prepared from 3-butyn-1-ol in five steps via protection, hydroxymethylation, hydroalumination—

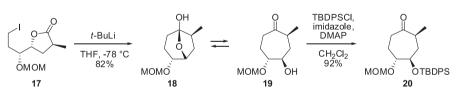
protonation, and mesylation—iodination. The diastereoselective alkylation of Evans' oxazolidinone **11** with **12** provided **13** in good yield with excellent stereoselectivity.¹¹ The stereocontrolled dihydroxylation of **13** with AD-mix- $\beta^{\otimes 12}$ was accompanied by spontaneous lactonization of the intermediate diol to afford the lactone **14** as a single isomer. Protection of the secondary alcohol with MOMCl and deprotection of the TBDPS ether with TBAF in THF,¹³ followed by mesylation—iodination, furnished the iodolactone **17** in good yield (Scheme 2).



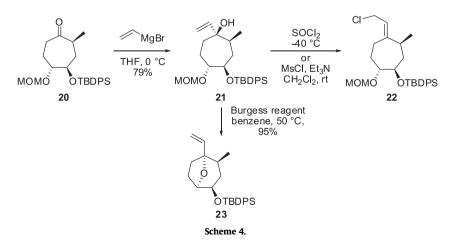
Scheme 2. Asymmetric synthesis of the γ -lactone 17.

The pivotal intramolecular acylation forming the seven-membered ring was first attempted with Sml₂ under various conditions and resulted in no reaction. Instead, lithiation of the iodide **17** using *t*-BuLi successfully provided the seven-membered ring in excellent yield.¹³ The NMR spectra showed that the product is in equilibrium between the hemiacetal **18** and the cycloheptanone **19** in CDCl₃. Treatment of the product with TBDPSCl and imidazole gave the trisubstituted cycloheptanone **20** in good yield (Scheme 3).

With the seven-membered skeleton in hand, we next attempted the introduction of the vinyl unit along with the formation of the C1–C5 endocyclic-olefin (Scheme 4). The vinyl Grignard reagent



Scheme 3. Intramolecular acylation of the alkyllithium to give the cycloheptanone.



added stereoselectively to the ketone to give **21** in good yield as a single isomer. According to the conformational analysis of **20**,¹⁴ the nucleophile would attack the α -face as an equatorial attack (Fig. 2). Attempts to dehydrate with thionyl chloride gave the chlorinated compound **22** via an S_N2' reaction, and the Burgess reagent resulted in formation of the oxabicyclic compound **23**.



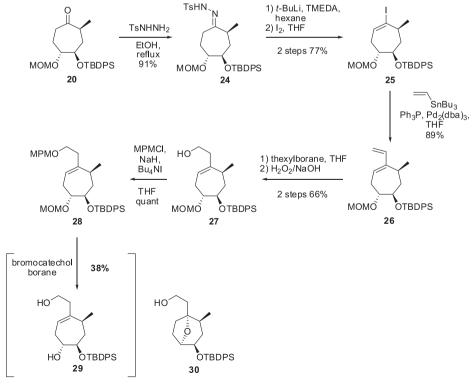
Fig. 2. Calculated conformation of 20 (TBDPS group was replaced by TMS for simplification).

We then tried the enol triflation of the ketone **20**, which gave only recovered starting material, but the Shapiro reaction of the corresponding *p*-toluenesulfonyl hydrazone **24** afforded the alkenyl iodide **25** in 77% yield after trapping of the alkenyllithium with iodine (Scheme 5). Stille coupling with **25** and tributylvinyltin provided the vinylcycloheptene **26**. Since we first intended to achieve the synthesis of sundiversifolide, the vinyl compound **26** was subjected to hydroboration with thexylborane to provide primary alcohol **27** after oxidative treatment. After protection of the hydroxyl group, deprotection of the MOM ether was attempted by using various kinds of acid, but resulted in low yield, and the oxabicyclic compound **30** was again obtained. The tertiary carbocation at C-1 would be easily generated to form this product.

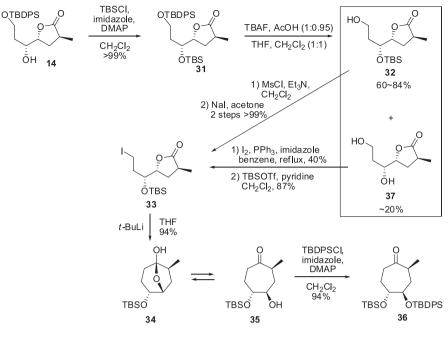
In place of the MOM ether, a TBS ether was selected as the protecting group of the hydroxyl group at C-7. Treatment of 14 with TBSCl gave 31, the primary TBDPS ether of which should be selectively deprotected. Although the usual protocol (TBAF, AcOH in THF)¹⁵ gave mainly the diol **37**. in the presence of CH₂Cl₂ as a cosolvent,¹⁶ the desired partially deprotected alcohol **32** was obtained in good yield. However, on a scale larger than 10 g of substrate, the yield of 32 decreased to 60% and more of the diol 37 was generated. The alcohol 32 was converted into the iodide 33 quantitatively. The diol 37 also can be transformed into the iodide 33 via selective iodination followed by protection of the secondary alcohol with TBSOTf. The iodide **33** was treated with *t*-BuLi to afford the hemiacetal **34** and the cycloheptanone **35** in excellent yield. Although MOMCl reacted with the hydroxyl groups on both 34 and 35, TBDPSCl protected only the secondary alcohol of 35 to afford 36 in good yield (Scheme 6).

The Shapiro reaction was carried out on the corresponding hydrazone **37** to give the doubly iodinated compound **38** in low yield after treatment with iodine (Scheme 7).

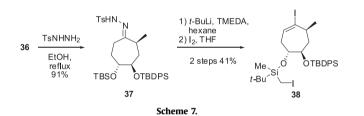
Hence, we then attempted the nucleophilic vinylation of **36** with a Grignard reagent as in the previous case.¹⁷ Vinylmagnesium bromide added to the ketone to afford the adduct **39** in excellent yield as a single isomer. Since dehydration of the tertiary alcohol of **39** to afford the endocyclic-olefin was unsuccessful at this stage as shown in Scheme 8, we decided to assemble the lactone first. Selective deprotection of the TBS group of **39**, TPAP oxidation of **40**, and removal of the TBDPS group of **41** provided the hydroxy hemiacetal **42** in nearly quantitative yield. We next tried to construct the lactone moiety from **42**, which is equivalent to an α -hydroxyketone. In order to carry out the intramolecular Horner–Emmons reaction, condensation with the carboxylic acid of the phosphonate ((EtO)₂P(O)CH(CH₃)CO₂H) and **42** gave a 1:1 mixture of **43** and **44**. On the contrary, reaction of **42** with the



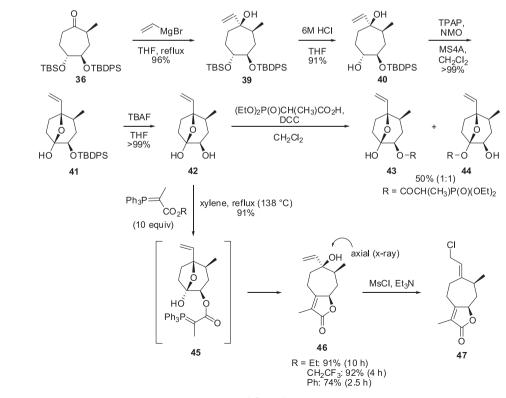
Scheme 5.







Wittig reagent (Ph₃P=C(CH₃)CO₂C₂H₅) under reflux in xylene for 9 h directly provided the butenolide **46** in 91% yield, the structure of which was unambiguously assigned by X-ray diffraction study. In contrast to Garner's report on the Wittig reaction of α -hydroxy ketones giving *E*-olefins,¹⁸ this reaction was exclusively *Z*-selective. The mechanism of this direct lactonization could be postulated as transesterification giving the intermediate **45**, followed by intramolecular olefination, because the ketone in **42** was masked by hemiacetalization. The reaction using the Wittig reagents with the trifluoroethyl ester (R=CH₂CF₃) and the phenyl ester (R=Ph), which

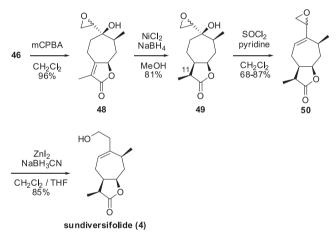


Scheme 8.

have better leaving groups, was completed in only 4 h (92% yield) and 2.5 h (74% yield), also suggesting that the intramolecular ole-fination had occurred (Scheme 8).

With the carbon skeleton in hand, we then investigated the less substituted endocyclic alkene formation. Dehydration using MsCl/ Et_3N gave the S_N2' products **47** and not the desired compound.

Since the double bond on the allylic alcohol had to be protected prior to dehydration. **46** was treated with *m*-CPBA to give the epoxide 48, which was subjected to hydrogenation with the nickel boride reagent derived from NiCl₂ and NaBH₄¹⁹ to afford a separable mixture of 49 (81%) and its C11-epimer (14%). After numerous attempts, the best results for a kinetically controlled dehydration of 49 were obtained using a protocol involving slow addition of a solution of freshly distilled SOCl₂ (2 equiv) and pyridine (4 equiv) in CH₂Cl₂ at -20 °C. Under these conditions, the desired endocyclic alkene **50** was successfully obtained in yields ranging of 68-87%. Finally, the epoxide in 50 was regioselectively reduced with NaBH₃CN in the presence of ZnI_2^{20} to provide sundiversifolide (**4**) in 85% yield. The synthetic **4** { $[\alpha]_{D}^{22}$ +33.0 (*c* 0.44, CHCl₃)} was found to be identical to the natural 4 according to spectroscopic properties (¹H and ¹³C NMR, IR, and MS) (Scheme 9). Since the optical rotation data of the natural product has not been reported due to a limited amount of the natural product, the enantiomer *ent*-1 { $[\alpha]_{D}^{22}$ -33.7 (*c* 0.46, CHCl₃)} was synthesized in a similar manner, using the chiral



Scheme 9. Synthesis of sundiversifolide (4).

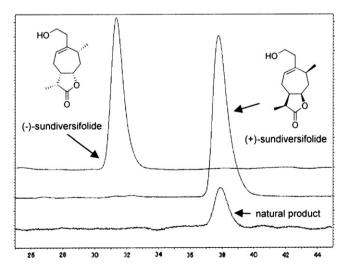


Fig. 3. HPLC analysis of natural and unnatural sundiversifolide using a chiral column (Daicel Chiralpak IA, detector: UV 207 nm, hexane/isopropanol 93:7, flow=1.0 mL/min; (–)-sundiversifolide t_r =31 min; (+)-sundiversifolide t_r =38 min, 30 °C).

oxazolidinone derived from L-Phe in the asymmetric alkylation and AD-mix- α^{\otimes} in the asymmetric dihydroxylation, which also gave a single isomer. HPLC analysis using chiral column (Daicel Chiralpak IA, detector: UV 207 nm, hexane/isopropanol 93:7, flow=1.0 mL/min; (-)-sundiversifolide t_r =31 min; (+)-sundiversifolide t_r =38 min, 30 °C) indicated that the natural **4** is identical with the synthetic (+)-**4** (Fig. 3). This study therefore determined that the absolute configuration of the naturally occurring sundiversifolide isolated from the sunflower is that shown in **1**.

2.2. Synthesis of (+)-8-epi-xanthatin (1)

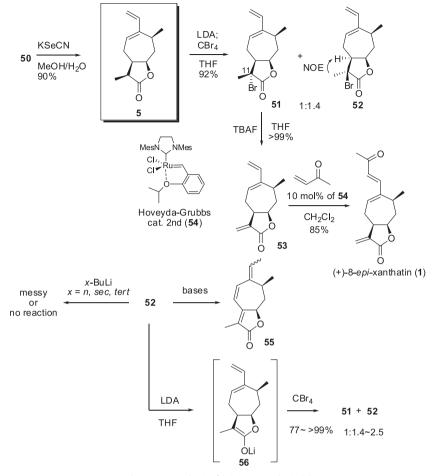
The key intermediate 5 for construction of xanthanolide has a *cis*-fused cycloheptene- γ -butyrolactone skeleton bearing a vinyl substituent, which has been converted into the butenone moiety via cross metathesis.⁸ Attempts to reduce the epoxide in **50** to an alkene using SmI_2^{21} or $\text{P}_2\text{I}_4^{22}$ resulted in a complex mixture, but KSeCN successfully reduced the epoxide to give **5** in good yield (Scheme 10).²³ With the key intermediate in hand, we tried to convert **5** to xanthanolides. In order to prepare 8-*epi*-xanthatin (**1**) from **5**, conversion of the α -methyl group on the γ -butyrolactone into an exo-methylene was examined. According to Ando's protocol,²⁴ bromination at C-11 was employed with LDA/CBr₄ to give a 1:1.4 diastereomeric mixture of 51 and 52, the stereochemistry of which was determined by NOE experiments as shown in Scheme 10. After separation of these diastereomers, each isomer was subiected to dehvdrobromination with base. The α -bromo isomer **51** was easily converted into the desired *exo*-olefin **53** by using TBAF in THF. The β -bromo isomer **52**, on the other hand, after being treated with various bases (TBAF, KHMDS, DBU, or Triton B) only gave the conjugated endo-olefin 55, probably due to the anti-periplanar relationship between the bromide and the β-proton on C-11. Thus, for isomerization of 52 to 51, formation of the enolate of 52 via lithium-halogen exchange, followed by bromination, was attempted using *n*-, sec- or tert-butyllithium, but the desired bromo lactone **51** was not obtained. Unexpectedly, LDA worked well to generate the enolate **56** via lithium-halogen exchange, affording a 1:1.4–1:2.5 mixture of 51 and 52 after bromination by CBr₄. Finally, 53 was subjected to cross metathesis⁸ using the second generation Hoveyda–Grubbs catalyst (54)²⁵ to give 8-epi-xanthatin (1) in good yield. The spectra of the synthetic 1 were identical with those of the natural compound.^{8a}

2.3. Synthesis of (-)-dihydroxanthatin (2)

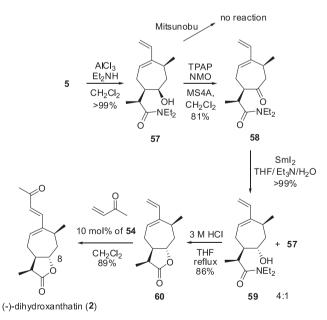
In order to synthesize dihydroxanthatin (2) from the intermediate **5**, a stereochemical inversion at C-8 was necessary. As we reported in the synthesis of diversifolide,^{9d} the lactone moiety was opened by diethylamine and aluminum trichloride to give the hydroxy amide **57** quantitatively. Since the inversion at C-8 by the Mitsunobu reaction was unsuccessful, probably due to steric reasons,²⁶ the oxidation–reduction protocol was examined. TPAP oxidation of the secondary alcohol at C-8, followed by thermodynamically controlled reduction with Sml₂,²⁷ provided a 4:1 mixture of the desired *trans*-isomer **59** and the *cis*-isomer **59** was cyclized by acid to furnish the *trans*-fused lactone **60**, which was subjected to cross metathesis with methyl vinyl ketone as described previously to yield (–)-dihydroxanthatin (**2**). All the spectra were identical with those of the reported compound^{8c} (Scheme 11).

2.4. Synthesis of (-)-xanthatin (3)

Xanthatin (**3**) could be synthesized by 'dehydrogenation' of the methyl moiety on dihydroxanthatin (**2**). According to the procedure described above, the precursor **60** was deprotonated by



Scheme 10. Synthesis of (+)-8-epi-xanthatin (1).



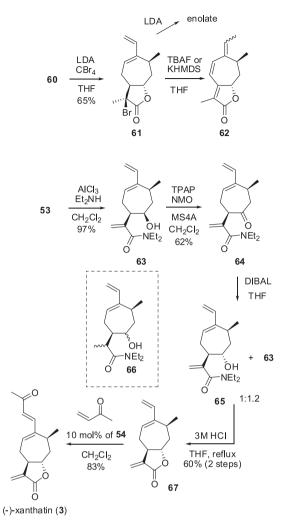
Scheme 11. Synthesis of (-)-dihydroxanthatin (2).

LDA, followed by bromination with CBr_4 , to give selectively the brominated compound **61** with the undesired stereochemistry.²⁸ The elimination reaction with base (TBAF, KHMDS) resulted in the formation of the *endo*-olefin **62**, as was seen with **52** (Scheme 10).

LDA induced lithiation to give the lithium enolate. Since the transfused lactone would have a planar structure, bromination would occur from the β -face of the enolate rather than the α -face to avoid steric hindrance with H-7. We then decided to carry out the stereochemical inversion of C-8 in 53, since it already had the exomethylene on the lactone. According to the stereochemical inversion procedure described above, the lactone was opened to give the hydroxy amide **63**, which was oxidized by TPAP to afford the ketone **64**. This compound was treated with SmI₂ in the same way, but gave 66, resulting from reduction of the enoate as well as the ketone. After screening several reducing reagents, we found that DIBAL reduced only the ketone to give a 1:1.2 ratio of the epimers (65:63). After lactonization with acid, cross metathesis was employed to provide (-)-xanthatin (3) (Scheme 12). The spectra of the synthetic 3 were identical with those of the reports.29

2.5. Conclusion

We have achieved the total synthesis of 8-*epi*-xanthatin (1), dihydroxanthatin (2), and xanthatin (3) starting from a common intermediate, prepared according to our original method, including an intramolecular acylation and a one-pot Wittig-lactonization, used in the synthesis of sundiversifolide. Since these xanthanolides have shown promising bioactivity, this work would be useful for the preparation of these compounds and the development of novel bioactive compounds derived from the natural products.



Scheme 12. Synthesis of (-)-xanthatin (3).

3. Experimental

3.1. General procedures

¹H NMR and ¹³C NMR spectra were measured in a CDCl₃ solution using JEOL INM AL-400 (¹H NMR at 400 MHz, ¹³C NMR at 100 MHz), and a JNM ECA-600 spectrometer (¹H NMR at 600 MHz, ¹³C NMR at 150 MHz) as the reference standards (¹H NMR at 0.00 ppm (TMS), ¹³C NMR at 77.0 ppm (CDCl₃)) unless otherwise noted. Chemical shifts are reported in parts per million. Peak multiplicities used the following abbreviation: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; br, broad. IR spectra were recorded on JASCO FT/IR-410 and Shimazu FT/IR-8300 spectrometers. Mass spectra and high resolution mass spectra were obtained on JMS-K9, JMS-AMSUN200/ 300, JMS-SX102A, JMS-DX303, and Waters LCT Premier mass spectrometers. Elemental analyses were performed with a Yanaco MT-3, MT-5, MT-6 CHN-Corde. Melting points were measured with a Yanaco MP-500D apparatus and a Seki Technotron Corp. Opti Melt MPA 100 apparatus and are uncorrected. Optical rotations were recorded on JASCO P-1010 and HORIBA SEPA-200. Analytical TLC was performed on precoated plates (0.25 mm, silica gel Merck 60 F_{2.54}). Column chromatography was performed on silica gel (Kanto Chemical Co., Inc.). Preparative HPLC was performed on Kanto Mightysil Si60 and Merck LiChrosorb Si60, and performed with a gradient solvent system of hexane and ethyl acetate and a UV detector at 254 nm. All reactions were performed under argon or nitrogen atmosphere unless otherwise noted, and dichloromethane (CH₂Cl₂), diethyl ether (Et₂O), and tetrahydrofuran (THF) were purchased from Kanto Chemical Co., Inc., and the other solvents were distilled. Unless otherwise noted, reagents were obtained from chemical sources and used without further purification.

3.2. Synthesis of sundiversifolide (4)

3.2.1. (E)-5-(tert-Butyldiphenylsiloxy)-1-iodo-2-pentene 12. To a solution of 5-(tert-butyldiphenylsiloxy)pent-2-en-1-ol (5.00 g, 14.7 mmol)³⁰ in CH₂Cl₂ (150 mL) at 0 °C were added Et₃N (5.1 mL, 36.7 mmol) and MsCl (2.3 mL, 29.4 mmol). The mixture was stirred for 0.5 h at room temperature, and then cooled to 0 °C. After addition of H₂O, the resulting mixture was extracted with CH₂Cl₂, washed with 5% HCl, saturated aqueous NaHCO₃, and brine, and dried over MgSO₄. The crude product was dissolved in acetone (150 mL) and NaI (6.60 g, 44.0 mmol) was added. The resulting suspension was stirred for 0.5 h at room temperature, and the solvent was removed in vacuo. After addition of H₂O, the mixture was extracted with CH₂Cl₂, washed with saturated aqueous Na₂S₂O₄ and brine, and dried over Na₂SO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (10% EtOAc/hexane) to give 12 as a yellow oil (6.53 g, 99%). ¹H NMR (400 MHz, CDCl₃) δ: 1.06 (s, 9H), 2.28 (dt, *J*=6.6, 6.6 Hz, 2H), 3.70 (t, J=6.6 Hz, 2H), 3.86 (d, J=7.4 Hz, 2H), 5.72 (dt, J=6.6, 15.2 Hz, 1H), 5.78 (dt, J=7.4, 15.2 Hz, 1H), 7.47 (m, 6H), 7.70 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ: 6.4 (t), 19.3 (s), 26.9 (q), 35.4 (t), 63.1 (t), 127.5 (d), 129.5 (d), 129.7 (d), 131.5 (d), 133.7 (s), 135.5 (d). IR (KBr, neat): 3070, 3048, 2955, 2930, 2857, 1471, 1427 cm⁻¹; MS (FAB) m/z; 473 (M^++Na) , 135 (100%); HRMS (FAB) calcd for C₂₁H₂₇IOSiNa (M⁺+Na) 473.0774, found: 473.0784.

3.2.2. (4R,2'S)-4-Benzyl-3-[7'-(tert-butyldiphenylsiloxy)-2'-methylhept-4'-enoyl]oxazolidin-2-one 13. To a solution of diisopropylamine (1.08 mL, 7.73 mmol) in THF (21 mL) at -78 °C was added BuLi in pentane (1.55 M, 4.9 mL, 7.73 mmol). The mixture was stirred for 20 min at -78 °C, and a solution of **11** (1.50 g, 6.44 mmol) in THF (10 mL) was added. After the reaction stirred for 1 h at -78 °C, a solution of 6 (4.35 g, 9.66 mmol) in THF (10 mL) was added to the mixture, which was stirred for 6 h at -40 °C. After addition of saturated aqueous NH₄Cl (3 mL), the resulting mixture was extracted with EtOAc. The combined organic layers were washed with brine (10 mL), and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (30% EtOAc/hexane) to give 13 as a colorless oil (3.29 g, 92%, >99%de). $[\alpha]_D^{26.4} - 20.1 (c 1.07, CHCl_3), enantiomer; [\alpha]_D^{20} + 20.37 (c 1.08, CHCl_3);$ ¹H NMR (400 MHz, CDCl₃) δ: 1.05 (s, 9H), 1.17 (d, *J*=6.8 Hz, 3H), 2.18 (ddd, J=7.2, 7.2, 12.8 Hz, 1H), 2.28 (dt, J=6.4, 6.4 Hz, 2H), 2.48 (ddd, *J*=6.4, 6.4, 12.8 Hz, 1H), 2.65 (dd, *J*=10.4, 12.8 Hz, 1H), 3.26 (dd, *J*=2.8, 12.8 Hz, 1H), 3.69 (t, J=7.2 Hz, 2H), 3.81 (tq, J=6.8, 6.8 Hz, 1H), 4.13 (dd, J=3.2, 7.6, 10.4 Hz, 1H), 5.47 (dt, J=6.4, 14.8 Hz, 1H), 5.54 (ddd, *I*=7.2, 6.4, 14.8 Hz, 1H), 7.20–7.40 (m, 11H), 7.64–7.70 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ: 16.4 (q), 19.2 (s), 26.9 (q), 36.6 (t), 36.9 (t), 37.6 (d), 38.1 (t), 55.3 (d), 63.8 (t), 66.0 (t), 127.2 (d), 127.5 (d), 127.6 (d), 128.8 (d), 129.3 (d), 129.4 (d), 133.9 (s), 135.4 (s), 135.5 (d), 152.9 (s), 176.5 (s). IR (KBr, neat): 2931, 2858, 1781, 1698 cm⁻¹; MS (FAB) m/z: 578 (M⁺+Na), 135 (100%); HRMS (FAB) calcd for C₁₃H₄₃NO₄SiNa (M⁺+Na) 556.2883, found: 556.2888.

3.2.3. (3S,5R,1'R)-5-[3'(tert-Butyldiphenylsiloxy)-1'-hydroxypropyl-3-methyldihydrofuran-2-one **14**. To a solution of AD-mix $\beta^{\text{(B)}}$ (16.33 g) in 50% t-BuOH/H₂O (30 mL) was added MeSO₂NH₂ (1.00 g, 10.4 mmol), and the mixture was stirred for 0.5 h at room temperature. The resulting mixture was cooled to 0 °C, and a solution of **13** (2.33 g, 4.17 mmol) in 50% t-BuOH/H₂O (30 mL) was added. The mixture was stirred for 10 h at 0 °C, and Na₂SO₃ (3.15 g, 25.06 mmol) was added. The resulting mixture was stirred for 0.5 h at room temperature, extracted with CHCl₃ (100 mL), and the combined organic layer was washed with satd NaHCO₃ aq (50 mL), brine (50 mL), and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (30% EtOAc/hexane) to give **14** (1.67 g, 97%): colorless needles (benzene/hexane): mp 93 °C. [α]_D²⁶ -33.19 (*c* 0.97, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ : 1.05 (s, 9H), 1.27 (d, *J*=7.2 Hz, 3H), 1.68 (ddt, *J*=3.2, 3.2, 14.8 Hz, 1H), 1.94 (m, 1H), 1.96 (m, 1H), 2.44 (ddd, *J*=3.6, 9.6, 12.4 Hz, 1H), 2.89 (td, *J*=7.2, 12.4 Hz, 1H), 3.94 (m, 1H), 4.39 (dt, *J*=3.6, 8.4 Hz, 1H), 7.26-7.47 (m, 6H), 7.66-7.68 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ : 16.3 (q), 19.1 (s), 26.8 (q), 32.5 (t), 34.1 (t), 62.5 (t), 73.1 (d), 80.2 (d), 127.2 (d), 127.7 (d), 129.7 (d), 132.6 (s), 132.9 (s), 135.4 (d), 180.4 (s). IR (CHCl₃): 3459, 1769 cm⁻¹; MS (FAB) *m/z*: 435 (M⁺+Na, 100%). Anal. Calcd for C₂₄H₃₂O₄Si: C, 69.86; H, 7.82. Found: C, 69.72; H, 7.85.

3.2.4. (3S,5R,1'R)-5-[1-(tert-Butyldimethylsiloxy)-3-(tert-butyldiphenylsiloxy)propyl]-3-methyldihydrofuran-2-one 31. To a solution of 14 (1.37 g, 3.33 mmol) in CH₂Cl₂ (6.7 mL) were added imidazole (793 mg, 11.7 mmol), 4-DMAP (40.3 mg, 0.33 mmol), and TBSCl (1.50 g, 9.99 mmol) at room temperature. The mixture was heated for 10 h under reflux. After addition of water, the resulting mixture was extracted with CH₂Cl₂, and the combined organic layers were washed with brine, and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (10-30% EtOAc/hexane) to give **31** as a colorless oil (1.75 g, quant.). $[\alpha]_D^{26}$ –15.6 (c 1.16, CHCl₃), enantiomer; $[\alpha]_D^{20}$ +13.09 (*c* 0.84, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 0.05 (s, 3H), 0.09 (s, 3H), 0.86 (s, 9H), 1.05 (s, 9H), 1.24 (d, *J*=7.2 Hz, 3H), 1.66-1.74 (m, 1H), 1.78-1.89 (m, 2H), 2.21 (ddd, J=4.4, 9.2, 12.8 Hz, 1H), 2.65–2.75 (m, 1H), 3.76 (t, *J*=6.0 Hz, 2H), 3.92 (dt, *J*=4.0, 6.4 Hz, 1H), 4.43 (dt, *J*=4.4, 8.4 Hz, 1H), 7.36-7.46 (m, 6H), 7.64 (dd, J=1.2, 6.4 Hz, 4H); ¹³C NMR (100 MHz, CDCl₃) δ : -4.5 (q), -4.4 (q), 16.6 (q), 18.1 (s), 19.2 (s), 25.9 (q), 26.9 (q), 32.2 (t), 34.3 (d), 35.7 (t), 60.1 (t), 71.5 (d), 79.1 (d), 127.6 (d), 129.6 (d), 129.7 (d), 133.4 (s), 133.5 (s), 135.4 (d), 180.1 (s). IR (KBr, neat): 1779 cm⁻¹. MS (EI) *m*/*z*: 526 (M⁺), 135 (100%); HRMS (EI) calcd for $C_{30}H_{46}O_4Si_2$ (M⁺) 526.2935, found: 526.2915.

3.2.5. (3S,5R,1'R)-5-[1-(tert-Butyldimethylsiloxy)-3-hydroxypropyl]-3-methyldihydrofuran-2-one 32. TBAF (1.0 M in THF, 0.02 mL, 0.02 mmol) and AcOH (0.01 mL, 0.019 mmol) were added to a solution of **31** (10 mg, 0.02 mmol) in THF (0.1 mL) and CH₂Cl₂ (0.1 mL). The resulting mixture was refluxed for 6 h, and saturated aqueous NaHCO3 was added to neutralize the mixture to a pH of 7. The aqueous phase was saturated with NaCl and extracted with EtOAc. The combined organic layers were dried over MgSO₄. The solvent was removed in vacuo and the residue was purified by column chromatography (30-75% EtOAc/hexane) to give 32 as a colorless oil (4.6 mg, 84%). $[\alpha]_{D}^{26} - 23.7 (c 1.00, CHCl_3)$, enantiomer; $[\alpha]_{D}^{20} + 17.24 (c - 1.00)$ 1.45, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 0.11 (s, 3H), 0.12 (s, 3H), 0.90 (s, 9H), 1.28 (d, J=7.2 Hz, 3H), 1.68-1.76 (m, 2H), 1.84-1.97 (m, 2H), 2.28 (ddd, J=4.4, 9.6, 13.2 Hz, 1H), 2.68-2.78 (m, 1H), 3.73-3.83 $(m, 2H), 3.94 (dt, J=6.4, 10.8 Hz, 1H), 4.55 (dt, J=4.4, 8.8 Hz, 1H); {}^{13}C$ NMR (100 MHz, CDCl₃) δ : -4.6 (q), -4.5 (q), 16.5 (q), 18.0 (s), 25.8 (q), 32.0 (t), 34.3 (d), 35.4 (t), 58.8 (t), 71.9 (d), 79.4 (d), 180.1 (s). IR (KBr, neat) 3444, 1770 cm⁻¹. MS (EI) m/z: 288 (M⁺), 157 (100%); HRMS (EI) calcd for C₁₄H₂₈O₄Si (M⁺) 288.1757, found: 288.1744.

3.2.6. (3S,5R,1'R)-5-[1-(tert-Butyldimethylsiloxy)-3-iodopropyl]-3methyldihydrofuran-2-one **33**. To a solution of **32** (3.70 g, 12.8 mmol) in CH₂Cl₂ (128 mL) were added Et₃N (3.6 mL, 25.6 mmol) and MsCl (1.9 mL, 19.2 mmol). The mixture was stirred for 30 min at room temperature, and then cooled to 0 °C. Water was added to the resulting mixture, which was diluted with CH₂Cl₂. The organic layer was washed with 5% aqueous HCl and saturated aqueous NaHCO₃, and was dried over MgSO₄. The solvent was removed in vacuo and the residue was dissolved in acetone (128 mL). NaI (11.5 g, 76.8 mmol) was added to the resulting mixture, which was stirred for 0.5 h at room temperature. The solvent was removed and water was added. The resulting mixture was extracted with CH₂Cl₂, and the combined organic layers were washed with saturated aqueous Na₂S₂O₄ and brine, then dried over MgSO₄. After removal of the solvent in vacuo, the residue was purified by column chromatography (10% EtOAc/hexane) to give a colorless oil 33 (5.10 mg, quant.). $[\alpha]_{D}^{26}$ –18.1 (c 1.06, CHCl₃), enantiomer; $[\alpha]_{D}^{20}$ +18.62 (c 1.02, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 0.13 (s, 6H), 0.89 (s, 9H), 1.28 (d, J=7.6 Hz, 3H), 1.90-2.01 (m, 2H), 2.08-2.16 (m, 1H), 2.30 (ddd, /=4.4, 9.6, 12.8 Hz, 1H), 2.68-2.78 (m, 1H), 3.19-3.28 (m, 2H), 3.85 (dt, J=4.4, 6.8 Hz, 1H), 4.47 (dt, J=4.0, 8.0 Hz, 1H); 13 C NMR (100 MHz, CDCl₃) δ : -4.4 (q), -4.3 (q), 1.7 (t), 16.5 (q), 18.0 (s), 25.8 (q), 31.8 (t), 34.1 (d), 36.4 (t), 74.0 (d), 78.3 (d), 179.6 (s). IR (KBr, neat): 1774 cm⁻¹. MS (EI) *m/z*: 398 (M⁺), 283 (100%); HRMS (EI) calcd for C₁₄H₂₇O₃SiI (M⁺) 398.0774, found: 398.0780.

3.2.7. (1R,4R,5R,7S)-4-(tert-Butyldimethylsilyloxy)-7-methyl-8-oxa*bicyclo*[3.2.1]*octan*-1-*ol* **34**. To a solution of **33** (5.10 g, 12.8 mmol) in THF (213 mL) was added t-BuLi (1.45 M in pentane, 21.2 mL, 30.7 mmol) at -78 °C. The mixture was stirred for 30 min, and saturated aqueous NH₄Cl was added. The resulting mixture was extracted with EtOAc, and the combined organic layers were washed with brine and dried over MgSO₄. The solvent was removed in vacuo and the residue was purified by column chromatography (15% EtOAc/hexane) to give **34** (3.29 mg, 94%) as a colorless solid. ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta$: 0.03 (s, 1.5H), 0.04 (s, 1.5H), 0.155 (s, 1.5H), 0.12 (s, 1.5H), 0.86 (s, 4.5H), 0.91 (s, 4.5H), 1.00 (d, J=6.8 Hz, 1.5H), 1.10 (d, *J*=6.8 Hz, 1.5H), 1.43–1.52 (m, 2H), 1.69–1.85 (m, 2H), 1.94 (tt, *J*=3.2, 14.4 Hz, 1H), 2.08–2.15 (m, 0.5H), 2.34 (dd, J=9.2, 13.2 Hz, 0.5H), 2.36–2.43 (m, 0.5H), 2.49 (dd, J=3.6, 12.8, 0.5 Hz), 2.56 (s, 0.5H), 2.70 (s, 0.5H), 3.40 (dt, J=3.6, 10.4 Hz, 0.5H), 3.50 (ddd, J=3.2, 7.6, 11.2 Hz, 0.5H), 3.96–4.00 (m, 0.5H); 13 C NMR (100 MHz, CDCl₃) δ : -4.7 (q), -4.7 (q), -4.6 (q), -4.1 (q), 17.3 (q), 17.6 (q), 18.0 (s), 25.8 (q), 27.1 (t), 29.7 (t), 32.5 (t), 35.3 (t), 36.5 (t), 36.6 (d), 37.6 (t), 41.7 (d), 68.0 (d), 76.0 (d), 76.4 (d), 78.6 (d), 104.1 (s), 214.0 (s). IR (0.2 mm KBr, CHCl₃): 3590, 1701 cm⁻¹; MS (EI) *m*/*z*: 272 (M⁺), 215 (100%); HRMS (EI) calcd for C₁₄H₂₈O₃Si (M⁺) 272.1808, found: 398.0780.

3.2.8. 5-(tert-Butyldimethylsiloxy)-4-(tert-butyldiphenylsiloxy)-2*methylcycloheptanone* **36***.* To a solution of **34** (11.4 mg, 0.04 mmol) in CH₂Cl₂ (0.08 mL) were added imidazole (5.1 mg, 0.07 mmol), 4-DMAP (catalytic amount), and TBDPSCl (0.01 mL, 0.05 mmol) at room temperature. The mixture was stirred for 2 days, and after addition of H₂O, the resulting mixture was extracted with CH₂Cl₂. The combined organic layers were washed with brine and dried over MgSO₄. The solvent was removed in vacuo and the residue was purified by column chromatography (3% EtOAc/hexane) to give a colorless oil **36** (20 mg, 94%). $[\alpha]_D^{26}$ –0.75 (*c* 1.06, CHCl₃), enantiomer; $[\alpha]_D^{20}$ +0.96 (*c* 1.04, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: -0.31 (s, 3H), -0.20 (s, 3H), 0.77 (s, 9H), 1.05 (s, 9H), 1.15 (d, J=6.8 Hz, 3H), 1.59-1.69 (m, 2H), 1.98-2.06 (m, 1H), 2.28-2.35 (m, 2H), 2.51–2.60 (m, 1H), 2.72 (ddd, J=3.2, 11.6, 12.8 Hz, 1H), 3.51 (dt, J=2.4, 5.6 Hz, 1H), 3.94 (ddd, J=2.0, 4.8, 7.2 Hz, 1H), 7.35–7.47 (m, 6H), 7.65 (dd, J=1.2, 8.0 Hz, 4H); ¹³C NMR (100 MHz, $CDCl_3$) δ : -5.1 (q), -5.0 (q), 17.9 (q), 18.2 (s), 19.2 (s), 25.7 (q), 27.2 (q), 27.2 (t), 32.1 (t), 36.6 (t), 42.6 (d), 71.7 (d), 74.8 (d), 127.6 (d), 127.7 (d), 129.7 (d), 129.7 (d), 133.8 (s), 133.9 (s), 135.7 (d), 135.8 (d), 213.5 (s). IR (KBr, neat) 1711 cm⁻¹; MS (EI) *m/z*: 510(M⁺), 193 (100%); HRMS (EI) calcd for C₃₀H₄₆O₃Si₂ (M⁺) 510.2986, found: 510.3013.

3.2.9. (1S,2S,4R,5R)-5-(tert-Butyldimethylsilyloxy)-4-(tert-butyldiphenylsilyloxy)-2-methyl-1-vinylcycloheptanol **39**. To a 1.0 M THF solution of vinylmagnesium bromide (3.5 mL, 3.51 mmol) was added 36 (0.89 g, 1.75 mmol) in THF (6.0 mL), and the mixture was refluxed for 6 h. After cooling to room temperature, the mixture was poured into 0.3 M HCl at 0 °C. The mixture was extracted with EtOAc, and the combined organic layers were washed with saturated aqueous NaHCO₃, and brine, and dried over MgSO₄. The solvent was removed in vacuo and the residue was purified by column chromatography (5% EtOAc/hexane), to give 39 as a colorless oil (0.90 g, 96%). $[\alpha]_D^{26} - 12.58 (c 1.04, CHCl_3)$, enantiomer; $[\alpha]_D^{20} + 14.96$ $(c 1.47, CHCl_3)$; ¹H NMR (400 MHz, CDCl₃) δ : -0.23 (s, 3H), -0.12 (s, 3H), 0.80 (s, 9H), 1.09 (s, 9H), 1.41–1.52 (m, 2H), 1.65–1.75 (m, 3H), 1.87 (m, 1H), 2.17 (m, 1H), 2.91 (s, 1H), 3.62 (m, 1H), 3.86 (m, 1H), 4.95 (dd, *J*=1.6, 10.8 Hz, 1H), 5.17 (dd, *J*=1.6, 17.2 Hz, 1H), 5.77 (dd, *J*=10.8, 17.2 Hz, 1H), 7.34–7.45 (m, 6H), 7.67–7.71 (m, 4H); ¹³C NMR $(100 \text{ MHz}, \text{CDCl}_3) \delta$: -4.8 (q), -4.7 (q), 17.9 (q), 18.5 (s), 19.1 (s), 25.8 (q), 26.0 (t), 27.1 (q), 35.1 (t), 36.2 (t), 38.4 (d), 73.8 (d), 75.7 (s), 76.6 (d), 109.8 (t), 127.5 (d), 127.6 (d), 129.6 (d), 129.7 (d), 133.5 (s), 133.7 (s), 135.8 (d), 135.9 (d), 146.2 (d). IR (KBr, neat): 3464, 1704 cm⁻¹; MS (ESI) m/z: 561 (M⁺+Na); HRMS (ESI) calcd for C₃₂H₅₀O₃NaSi₂ (M⁺+Na) 561.3196, found: 561.3192.

3.2.10. (1S,4R,5R,7S)-5-(tert-Butyldiphenylsilyloxy)-7-methyl-1-vinylcycloheptane-1,4-diol 40. To a solution of 39 (1.35 g, 1.61 mmol) in THF (14.2 mL) was added 6 M HCl (1.8 mL). The mixture was stirred for 18 h at room temperature, and then extracted with EtOAc. The combined organic layers were washed with saturated aqueous NaHCO3 and brine, and dried over MgSO4. The solvent was removed in vacuo and the residue was purified by column chromatography (30% EtOAc/hexane) to give 40 as colorless needles (0.62 g, 91%): (EtOAc/hexane): mp 110 °C; $[\alpha]_D^{25}$ -49.55 (c 1.12, CHCl₃), enantiomer; $[\alpha]_D^{20}$ +53.19 (*c* 0.94, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 0.51 (d, *J*=6.8 Hz, 3H), 1.04 (s, 9H), 1.13 (m, 3H), 1.36 (m, 1H), 1.55–1.60 (m, 1H), 1.76–1.83 (m, 2H), 1.90 (m, 1H), 2.42 (s, 1H), 3.53 (m, 1H), 3.64 (m, 1H), 4.96 (dd, J=1.2, 11.2 Hz, 1H), 5.079 (dd, J=1.2, 17.2 Hz, 1H), 5.75 (dd, J=11.2, 17.2 Hz, 1H), 7.40 (m, 6H), 7.66 (m, 4H); ¹³C NMR (150 MHz, CDCl₃) δ: 16.1 (q), 19.3 (s), 25.3 (t), 27.0 (q), 35.5 (t), 35.7 (t), 36.5 (d), 75.5 (s), 78.5 (d), 81.1 (d), 110.9 (d), 127.6 (d), 127.8 (d), 129.7 (d), 129.8 (d), 133.7 (s), 135.6 (d), 135.7 (d), 145.3 (d). IR (0.1 mm NaCl, CHCl₃): 3427, 1633 cm⁻¹; MS (FAB) *m*/*z*: 425 (M⁺+1). Anal. Calcd for C₂₆H₃₆O₃Si: C, 73.54; H, 8.54. Found: C, 73.47; H, 8.49.

3.2.11. (1S,2R,4S,5S)-2-(tert-Butyldiphenylsilyloxy)-4-methyl-5-vinyl-8-oxabicyclo[3.2.1]octan-1-ol 41. To a solution of 40 (0.62 g, 1.46 mmol) in CH₂Cl₂ (10 mL) were added 4-methylmorpholine N-oxide (0.85 g, 7.30 mmol) and molecular sieves 4 Å (0.73 g). The mixture was stirred at room temperature for 10 min, and tetrapropylammonium perruthenate (0.02 g, 0.07 mmol) was added. The mixture was stirred for 1 h, and filtered through a pad of Celite. The solvent was removed from the filtrate and the crude product was purified with column chromatography (30% EtOAc/ hexane) to give **41** as a colorless oil (0.61 g, quant.). $[\alpha]_D^{25}$ –40.62 (*c* 1.35, CHCl₃), enantiomer; $[\alpha]_D^{20}$ +54.54 (*c* 0.33, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 1.09 (s, 9H), 1.13 (d, J=6.8 Hz, 3H), 1.46 (m, 2H), 1.69–1.95 (m, 5H), 3.72 (s, 1H), 3.77 (m, 1H), 5.04 (dd, J=1.48, 11 Hz, 1H), 5.23 (dd, *J*=1.48, 17.5 Hz, 1H), 5.93 (dd, *J*=11, 17.5 Hz, 1H), 7.40 (m, 6H), 7.70 (m, 4H); ¹³C NMR (150 MHz, CDCl₃) δ: 17.4 (q), 19.3 (s), 27.1 (q), 32.0 (t), 32.4 (t), 34.3 (t), 35.5 (d), 72.2 (d), 85.2 (s), 104.3 (s), 112.0 (t), 127.7 (d), 127.9 (d), 129.8 (d), 129.9 (d), 133.1 (s), 133.4 (d), 135.8 (d), 136.2 (d), 141.3 (d). IR (KBr neat):3551 cm⁻¹; MS (FAB) m/z: 422 (M⁺); HRMS (FAB) calcd for C₂₆H₃₄O₃Si (M⁺+Na) 422.2277, found: 422.2278.

3.2.12. (1S,2R,4S,5S)-4-Methyl-5-vinyl-8-oxabicyclo[3.2.1]octane-1,2-diol **42**. To a solution of **41** (0.16 g, 1.46 mmol) in THF (15 mL) was added TBAF (1.0 M in THF, 1.7 mL). The mixture was stirred at room temperature for 30 min, and the solvent was removed in vacuo. The residue was purified with column chromatography (30–60% EtOAc/hexane) to give **42** as colorless needles (0.26 g, quant.). Colorless needles (EtOAc/hexane): mp 63–64 °C; $[\alpha]_D^{25}$ –50.98 (*c* 1.02, CHCl₃), enantiomer; $[\alpha]_D^{20}$ +64.38 (*c* 0.73, CHCl₃); 1H NMR (400 MHz, CDCl₃) δ : 1.11 (d, *J*=7.2 Hz, 3H), 1.60 (m, 1H), 1.76 (m, 1H), 1.84–2.00 (m, 4H), 2.04 (m, 1H), 2.16 (m, 1H), 3.60 (m, 1H), 3.88 (s, 1H), 5.08 (dd, *J*=1.2, 11.2 Hz, 1H), 5.26 (dd, *J*=1.2, 18 Hz, 1H), 5.90 (dd, *J*=11.2, 18 Hz); ¹³C NMR (150 MHz, CDCl₃) δ : 18.1 (q), 32.1 (t), 32.9 (t), 34.4 (t), 35.1 (d), 70.9 (d), 85.4 (s), 104.5 (s), 112.1 (t), 140.5 (d). IR (0.1 mm NaCl, CHCl₃):3537 cm⁻¹; MS (FAB) *m/z*: 185 (M⁺+1), 154 (100%). Anal. Calcd for C₁₀H₁₆O₃: C, 65.19; H, 8.75. Found: C, 65.11; H, 8.72.

3.2.13. (6S,7S,8aR)-6-Hydroxy-3,7-dimethyl-6-vinyl-4,5,6,7,8,8a-hexahydro-2H-cyclohepta/b/furan-2-one 46. To a solution of 42 (0.10 g, 0.54 mmol) in xylene (10.8 mL) was added the Wittig reagent (Ph₃PC (CH₃)CO₂CH₂CF₃, 1.80 g, 4.32 mmol). The resulting mixture was refluxed for 4 h. The solvent was removed and the crude residue was purified with column chromatography (50% EtOAc/hexane) to give **46** as colorless prisms (0.112 g, 92%): (EtOAc/hexane): mp 151 °C; $[\alpha]_D^{25}$ –140.27 (c 1.08, CHCl₃), enantiomer; $[\alpha]_D^{20}$ +133.11 (c 1.54, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 0.93 (d, *J*=6.8 Hz, 3H), 1.61 (m, 1H), 1.775 (m, 1H), 1.779 (s, 3H), 1.86 (m, 2H), 2.11 (m, 1H), 2.53 (m, 1H), 2.89 (m, 1H), 4.87 (br d, J=12.4 Hz, 1H), 5.09 (d, J=10.8 Hz, 1H), 5.19 (d, J=17.6 Hz, 1H), 5.93 (dd, J=10.8, 17.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ: 8.2 (q), 17.1 (q), 20.8 (t), 36.0 (t), 36.3 (t), 39.9 (d), 75.3 (s), 82.1 (d), 110.4 (t), 120.8 (d), 146.3 (d), 164.8 (s), 174.5 (s). IR (CHCl₃): 3471, 1741, 1672 cm⁻¹. MS (EI) *m/z*: 222 (M⁺), 204 (100%); Anal. Calcd for C₁₃H₁₈O₃: C, 70.24; H, 8.16. Found: C, 70.05; H, 8.12. CCDC-629263 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/conts.retrieving.html (or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB21EZ, UK; fax: (+44)1223 336 033; or deposit@ccdc.cam.ac.uk).

3.2.14. (6R,7S,8aR)-6-Hydroxy-3,7-dimethyl-6-(oxiran-2-yl)-4,5,6,7,8,8a-hexahydro-2H-cyclohepta/b]furan-2-one 48. To a solution of 46 (50 mg, 0.22 mmol) in CH₂Cl₂ (2.2 mL) was added *m*-chloroperbenzoic acid (155.3 mg, 0.90 mmol). The mixture was stirred for 10 h at room temperature, and then saturated aqueous NaHCO3 and brine were added. The mixture was extracted with CH₂Cl₂, and the combined organic layers were washed with brine and dried over MgSO₄. The solvent was removed and the crude residue was purified with column chromatography (60% EtOAc/ hexane) to give 48 (50.2 mg, 96%) as colorless prisms; $(CH_2Cl_2/$ hexane); mp 220 °C; $[\alpha]_D^{22}$ –99.25 (*c* 1.35, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 1.01 (d, *J*=6.8 Hz, 3H), 1.39 (s, 1H), 1.67 (ddd, *J*=11.2, 12, 14 Hz, 1H), 1.77 (s, 3H), 1.81–1.89 (m, 2H), 2.59 (m, 1H), 2.81 (m, 1H), 2.91 (d, *J*=4 Hz, 2H), 2.98 (t, *J*=4 Hz, 1H), 4.89 (br d, J=11.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ : 8.3 (q), 16.7 (q), 20.7 (t), 34.0 (t), 36.7 (t), 38.5 (d), 46.5 (t), 58.0 (d), 71.3 (s), 81.9 (d), 121.1 (s), 164.4 (s), 174.4 (s). IR (CHCl₃): 3018, 1743 cm⁻¹. MS (EI) *m/z*: 238 (M⁺), 177 (100%); Anal. Calcd for C₁₃H₁₈O₄: C, 66.53; H, 7.61. Found: C, 65.26; H, 7.58.

3.2.15. (3S,3aR,6R,7S,8aR)-6-Hydroxy-3,7-dimethyl-6-(oxiran-2-yl) octahydro-2H-cyclohepta[b] furan-2-one **49**. To a suspension of **48** (139 mg, 0.58 mmol) and NiCl₂ (19 mg, 0.14 mmol) in MeOH (10 mL) was added NaBH₄ (88 mg, 2.34 mmol) at 0 °C. The mixture was stirred at 0 °C for 2 h, and then saturated aqueous NH₄Cl (5 mL) was added. The resulting mixture was diluted with CH₂Cl₂ and the aqueous layer was extracted with CH₂Cl₂. The combined organic layers were washed with brine and dried over MgSO₄. The solvent was removed and the crude residue was purified by HPLC (60% EtOAc/hexane) to give **49** as colorless prisms (114 mg, 81%, 4:1

mixture of diastereomers of the oxirane) and C11-isomer (14%). Major isomer of **49**: colorless prisms (CH₂Cl₂/hexane); mp 150 °C; $[\alpha]_D^{22}$ –5.00 (*c* 1.00, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ : 1.00 (d, *J*=7.2 Hz, 3H), 1.20 (d, *J*=7.2 Hz, 3H), 1.45–1.59 (m, 2H), 1.72 (m, 3H), 2.02 (ddd, *J*=7.6, 7.6, 8.0 Hz, 1H), 2.35 (ddd, *J*=10.4, 11.2, 14.8 Hz, 1H), 2.58 (m, 1H), 2.78–2.87 (m, 3H), 2.93 (dd, *J*=2.8, 5.2 Hz, 1H), 4.67 (ddd, *J*=6.6, 12 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ : 11.1 (q), 17.3 (t), 17.8 (q), 33.2 (t), 34.5 (d), 38.9 (d), 40.9 (t), 43.4 (d), 45.1 (t), 58.3 (d), 71.2 (s), 80.8 (r), 179.1 (s). IR (0.1 mm NaCl, CHCl₃): 1762 cm⁻¹; MS (FAB) *m/z*: 241 (M⁺+1). Anal. Calcd for C₁₃H₂₀O₄: C, 64.98; H, 8.39. Found: C, 64.93; H, 8.36.

3.2.16. (3S,3aR,7S,8aR)-3,7-Dimethyl-6-(oxiran-2-yl)-3,3a,4,7,8,8ahexahydro-2H-cyclohepta[b] furan-2-one 50. To a solution of 49 (3.0 mg, 0.0125 mmol) in CH₂Cl₂ (1.2 mL) was added dropwise a solution of freshly distilled thionyl chloride (1.8 µL, 0.025 mmol) and pyridine (4 $\mu L)$ in CH_2Cl_2 (0.3 mL) at -20 °C. The resulting mixture was stirred at -20 °C for 10 min then saturated aqueous NaHCO₃ was added. The resulting mixture was extracted with CH₂Cl₂, and the combined organic layer was washed with aqueous CuSO₄ and dried over MgSO₄. The solvent was removed in vacuo and the residue was purified by column chromatography (20% EtOAc/hexane) to give 50 as a colorless oil (2.5 mg, 87%); Major isomer: mp 71.5–72 °C (hexane-EtOAc); $[\alpha]_D^{22}$ –16.22 (*c* 0.31, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ: 1.16 (d, *J*=7.2 Hz, 3H), 1.20 (d, *J*=6.8 Hz, 3H), 1.95–2.19 (m, 4H), 2.47 (dd, *J*=6.4, 3.2 Hz, 1H), 2.48 (m, 1H), 2.73 (m, 1H), 2.83 (m, 1H), 2.91 (dd, *J*=6.4, 4.4 Hz, 1H), 3.23 (br s, 1H), 4.65 (m, 1H), 5.73 (br d, *J*=9.6 Hz, 1H); ¹³C NMR (100 MHz. CDCl₃) δ : 10.6 (q), 21.0 (q), 31.8 (t), 36.9 (d), 38.9 (d), 42.3 (t), 50.4 (d), 52.7 (d), 53.2 (d),80.1 (t), 121.4 (d), 141.2 (s), 178.8 (s). IR (NaCl, neat): 1770 cm⁻¹. MS (EI) m/z: 222 (M⁺), 60 (100%); HRMS (EI) calcd for C₁₃H₁₈O₃ (M⁺) 222.1256, found: 222.1258.

3.2.17. Sundiversifolide (4). To a solution of 50 (18.6 mg, 0.0837 mol) in CH₂Cl₂ (1.5 mL) and THF (0.1 mL) were added ZnI₂ (80.1 mg, 0.25 mmol) and NaBH₃CN (16.5 mg, 0.25 mmol). The mixture was stirred for 2 h at room temperature, and 1 M aqueous HCl (0.1 mL) was added at 0 °C. The resulting mixture was saturated with NaCl and extracted with CH₂Cl₂. The organic layer was dried over MgSO₄. The solvent was removed in vacuo and the residue was purified by column chromatography (70% EtOAc/hexane) to give 4 as a colorless solid (15.9 mg, 85%). Mp 42–47 °C; $[\alpha]_D^{20}$ +33.0 (c 0.44, CHCl₃), enantiomer; [α]²⁰_D –33.7 (*c* 0.46, CHCl₃); ¹H NMR (400 MHz, CD₃OD) δ: 1.11 (d, *J*=7.3 Hz, 3H), 1.15 (d, *J*=7.1 Hz, 3H), 1.87–1.99 (m, 2H), 2.01-2.04 (m, 2H), 2.10-2.23 (m, 2H), 2.72-2.80 (m, 1H), 2.86-2.94 (m, 1H), 3.50-3.63 (m, 2H), 4.67-4.73 (m, 1H), 5.53–5.55 (m, 1H); ¹³C NMR (150 MHz, CD₃OD) δ: 10.9, 21.8, 22.8, 33.9, 38.2, 40.1, 41.2, 43.5, 62.1, 82.5, 125.3, 143.6, 181.8. IR (neat): 3415, 1651, 1732 cm⁻¹; HRMS (EI) calcd for C₁₃H₂₀O₃ (M⁺) 224.1412, found: 224.1405.

3.3. Synthesis of 5

3.3.1. (3*S*,3*a*,7*S*,8*a*,)-3,7-*Dimethyl*-6-*vinyl*-3,3*a*,4,7,8,8*a*-*hexahydro-2H*-*cyclohepta*[*b*]*furan-2-one* **5**. To a solution of the epoxide **50** (6.2 mg, 27.9 µmol) in H₂O/methanol (1:10, 0.3 mL) was added potassium selenocyanate (16.1 mg, 111.6 µmol). The mixture was stirred for 10 h at room temperature. The reaction was filtered, diluted with water, and extracted with Et₂O. The combined organic layers were washed with brine and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (10% EtOAc/Hex) to give **5** (5.2 mg, 90%) as colorless prisms: mp 53.0–54.0 °C (EtOAc/Hex); $[\alpha]_D^{21}$ –21.21 (*c* 0.33, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.19 (d, *J*=6.6 Hz, 3H), 1.23 (d, *J*=4.8 Hz, 3H), 1.98–2.05 (m, 2H), 2.16 (ddd, *J*=13.8, 5.4, 3.6 Hz, 1H), 2.43–2.48 (m, 1H), 2.71–2.86 (m, 3H), 4.63 (ddd, *J*=11.4, 7.2, 3.6 Hz, 1H), 4.96 (d, *J*=11.4 Hz, 1H), 5.12 (d, *J*=11.4 Hz, 1H), 5.74 (dd, *J*=9.0, 5.4 Hz, 1H), 6.17 (dd, *J*=17.4, 11.4 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ : 12.2 (q), 21.7 (q), 22.2 (t), 31.1 (d), 36.7 (t), 38.1 (d), 40.5 (d), 80.5 (d), 111.9 (t), 127.1 (d), 139.8 (d), 144.2 (s), 179.6 (s). IR (KBr): 2933, 1743 cm⁻¹; MS (EI) *m/z*: 206 (M⁺), 91 (100%); HRMS (EI) calcd for C₁₃H₁₈O₂: 206.1307, found: 206.1311.

3.4. Synthesis of (+)-8-epi-xanthatin (1)

3.4.1. (3R,3aS,7S,8aR)-3-Bromo-3,7-dimethyl-6-vinyl-3,3a,4,7,8,8ahexahydro-2H-cyclohepta [b]furan-2-one **51**. To a solution of **5** (70.0 mg, 0.34 mmol) in THF (3.4 mL) was added LDA (0.61 M) (0.72 mL, 0.44 mmol) at -78 °C. The mixture was stirred for 1 h at -78 °C, and then CBr₄ (225 mg, 0.68 mmol) in THF (1.0 mL) was added dropwise at -78 °C. After 20 min at -78 °C, the reaction was quenched with aqueous saturated NH₄Cl, and the mixture was extracted with CH₂Cl₂. The combined organic layers were washed with brine and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (5% EtOAc/Hex) to give **51** (37.4 mg, 39%) as a colorless oil and **52** (53%) as a white solid.

Compound **51**: $[\alpha]_D^{25}$ –44.4 (*c* 0.27, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.19 (d, *J*=7.2 Hz, 3H), 1.87 (s, 3H), 2.00–2.06 (m, 1H), 2.15–2.24 (m, 2H), 2.37–2.42 (m, 1H), 2.67–2.73 (m, 1H), 3.10 (ddd, *J*=13.2, 7.2, 3.0 Hz, 1H), 4.94 (ddd, *J*=11.4, 6.6, 4.8 Hz, 1H), 4.99 (d, *J*=10.8 Hz, 1H), 5.13 (d, *J*=18.0 Hz, 1H), 5.73 (dd, *J*=9.0, 4.2 Hz, 1H), 6.17 (dd, *J*=18.0, 10.8 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ : 22.1 (q), 22.9 (t), 24.1 (q), 30.9 (d), 35.8 (t), 52.5 (d), 57.3 (s), 79.4 (d), 112.8 (t), 125.2 (d), 139.6 (d), 145.2 (s), 174.3 (s). IR (CHCl₃): 2958, 1774, 1207 cm⁻¹; MS (EI) *m/z*: 286 ([M+H]⁺), 284 ([M-1]⁺), 79 (100%); HRMS (EI) calcd for C₁₃H₁₇O₂Br: 284.0412, found: 284.0411.

Compound **52**: mp 86–87.5 °C; $[\alpha]_{D}^{25}$ –16.7 (*c* 0.18, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.19 (d, *J*=6.6 Hz, 3H), 1.93 (s, 3H), 2.16 (ddd, *J*=14.4, 7.8, 1.8 Hz, 1H), 2.25 (ddd, *J*=13.8, 9.0, 4.8 Hz, 1H), 2.43–2.49 (m, 2H), 2.79–2.88 (m, 2H), 4.51 (ddd, *J*=12.6, 8.4, 1.8 Hz, 1H), 5.00 (d, *J*=10.8 Hz, 1H), 5.15 (d, *J*=17.4 Hz, 1H), 5.74 (dd, *J*=9.0, 6.0 Hz, 1H), 6.16 (dd, *J*=17.4, 10.8 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ : 21.2 (q), 25.9 (t), 27.8 (q), 31.2 (d), 33.0 (t), 47.1 (d), 58.5 (s), 79.1 (d), 112.0 (t), 125.5 (d), 139.6 (d), 143.9 (s), 174.8 (s). IR (CHCl₃): 2941, 1762, 1236, 1209, 1105, 979 cm⁻¹; MS (EI) *m/z*: 286 ([M+H]⁺), 284 ([M–1]⁺), 91 (100%); HRMS (EI) calcd for C₁₃H₁₇O₂Br: 284.0412, found: 284.0417.

Conversion of **52** to **51**: To a solution of **52** (7.7 mg, 27.0 µmol) in THF (300 µL) was added LDA (0.61 M, 66 µL, 40.5 µmol) at $-78 \degree$ C. The mixture was stirred for 1 h at $-78 \degree$ C, then CBr₄ (17.9 mg, 54 µmol) in THF (200 µL) was added dropwise at $-78 \degree$ C. After 20 min at $-78 \degree$ C, the reaction was quenched with aqueous saturated NH₄Cl, and the mixture was extracted with CH₂Cl₂. The combined organic layers were washed with brine, then dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (5% EtOAc/Hex) to give **51** (3.3 mg, 43%) as a colorless oil and **52** (4.4 mg, 57%) as a white solid.

3.4.2. (3aR,7S,8aR)-7-Methyl-3-methylene-6-vinyl-3,3a,4,7,8,8ahexahydro-2H-cyclohepta[b] furan-2-one **53**. To a solution of **51** (3.9 mg, 13.7 µmol) in THF (0.35 mL) was added TBAF (1.0 M, 27.4 µL, 27.4 µmol) at room temperature. The reaction was stirred for 30 min at room temperature before the reaction was quenched with aqueous saturated NH₄Cl. The mixture was extracted with CH₂Cl₂, and the combined organic layers were washed with brine, and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (10% AcOEt/ Hex) to give **53** (2.8 mg) as colorless needles: mp 74.5–75.5 °C (EtOAc/Hex); $[\alpha]_D^{25}$ +14.8 (*c* 0.27, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.18 (d, *J*=6.6 Hz, 3H), 1.84–1.90 (m, 1H), 2.13 (ddd, *J*=13.8, 7.2, 2.4 Hz, 1H), 2.37 (ddd, *J*=13.8, 8.4, 4.8 Hz, 1H), 2.51 (ddd, *J*=13.8, 13.8, 6.0 Hz, 1H), 2.80–2.87 (m, 1H), 3.36–3.42 (m, 1H), 4.66 (ddd, J_1 =12.0, 8.4, 2.4 Hz, 1H), 5.00 (d, J=11.4 Hz, 1H), 5.14 (d, J=17.4 Hz, 1H), 5.54 (d, J=3.0 Hz, 1H), 5.77 (dd, J=9.0, 6.6 Hz, 1H), 6.16 (dd, J=17.4, 11.4 Hz, 1H), 6.29 (d, 1H, J=3.0 Hz); ¹³C NMR (150 MHz, CDCl₃) δ : 21.5 (q), 26.3 (t), 31.3 (d), 36.4 (t), 41.6 (d), 78.8 (d), 112.1 (t), 122.1 (d), 126.4 (s), 138.6 (d), 139.6 (d), 143.9 (s), 170.2 (s). IR (CHCl₃): 2931, 1747, 1272, 977 cm⁻¹; MS (EI) m/z: 204 (M⁺), 93 (100%); HRMS (EI) calcd for C₁₃H₁₆O₂: 204.1150, found: 204.1153.

3.4.3. (+)-8-epi-Xanthatine **1**. To a solution of **53** (1.7 mg, 8.3 μmol) in dry CH₂Cl₂ (1.5 mL), the second generation Hoveyda–Grubbs catalyst 54 (0.5 mg, 0.83 µmol) was added. Freshly distilled methyl vinyl ketone (13.4 µL, 166.4 µmol) was added to the mixture using a syringe pump over 8 h at 45 °C. Upon completion, the mixture was cooled to room temperature before DMSO (15.0 µL) was added. After 6 h at room temperature, the mixture was evaporated, and the residue was purified by silica gel column chromatography (30% EtOAc/Hex) to give 8-epi-xanthatin (1) (1.7 mg, 85%) as a colorless oil. $[\alpha]_D^{23}$ +44.0 (c 0.25, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.18 (d, J=6.6 Hz, 3H), 1.88–1.94 (m, 1H), 2.17 (ddd, J=13.8, 7.2, 2.4 Hz, 1H), 2.30 (s, 3H), 2.57–2.65 (m, 1H), 2.50 (ddd, J=13.8, 9.0, 4.8 Hz, 1H), 2.61 (ddd, J=13.8, 13.8, 6.0 Hz, 1H), 2.80-2.86 (m, 1H), 3.39-3.44 (m, 1H), 4.65 (ddd, J=12.0, 8.4, 2.4 Hz, 1H), 5.57 (d, J=2.4 Hz, 1H), 6.13 (d, J=16.2 Hz, 1H), 6.20 (dd, J=9.0, 6.0 Hz, 1H), 6.32 (d, J=3.6 Hz, 1H), 6.97 (d, *J*=16.2 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ: 21.5 (q), 27.0 (q), 27.7 (d), 31.7 (q), 36.3 (t), 41.1 (d), 78.2 (d), 122.5 (t), 125.8 (d), 135.7 (d), 138.1 (s), 142.9 (s), 146.4 (d), 169.8 (s), 198.5 (s). IR (CHCl₃): 3018, 1758, 1668, 1593, 1274, 1207 cm⁻¹; MS (FAB) m/z: 247 $([M+H]^+)$; HRMS (FAB) calcd for $C_{15}H_{19}O_3$ ($[M+H]^+$): 247.1334, found: 247.1337.

3.5. Synthesis of (-)-dihydroxanthatin (2)

3.5.1. (S)-N,N-Diethyl-2-((1R,5S,7R)-7-hydroxy-5-methyl-4-vinylcyclohept-3-enyl)propanamide 57. To a solution of 5 (20.0 mg, 97.9 μ mol) in CH₂Cl₂ (1.0 mL) were added AlCl₃ (39.1 mg, 293.7 μ mol) and Et₂NH (30.6 mL, 293.7 μ mol) at room temperature. After 30 min at room temperature, the reaction was quenched with 1 M HCl, whereupon the mixture was extracted with CH₂Cl₂. The combined organic layers were washed with saturated aqueous NaHCO₃ and brine, and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (50% AcOEt/Hex) to give 57 (27.4 mg) as a colorless oil. $[\alpha]_D^{23}$ –75.0 (*c* 0.40, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.12–1.20 (m, 6H), 1.19 (t, J=7.2 Hz, 3H), 1.28 (d, J=7.2 Hz, 3H), 1.72 (dd, J=15.0, 9.0 Hz, 1H), 1.82 (dt, J=15.0, 3.6 Hz, 1H), 1.86 (br s, 1H), 1.93-1.97 (m, 1H), 2.02 (dt, J=15.0, 5.4 Hz, 1H), 2.61 (ddd, J=15.6, 11.4, 4.8 Hz, 1H), 2.75-2.84 (m, 2H), 3.30-3.45 (m, 4H), 4.31 (br s, 1H), 4.87 (d, *J*=10.8 Hz, 1H), 5.07 (d, *J*=16.8 Hz, 1H), 5.78 (dd, *J*=9.0, 4.2 Hz, 1H), 6.21 (dd, *J*=16.8, 10.2 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ: 13.1 (q), 15.0 (q), 15.9 (q), 19.8 (t), 26.0 (t), 30.6 (d), 38.2 (d), 40.0 (t), 40.6 (t), 42.1 (t), 43.9 (q), 71.5 (d), 109.6 (d), 131.3 (d), 140.5 (d), 145.8 (s), 176.1 (s). IR (CHCl₃): 2985, 2358, 1731, 1618, 1462, 1434 cm⁻¹; MS (FAB) m/z: 280 ([M+H]⁺); 280 (100%); HRMS (FAB) calcd for C₁₇H₃₀NO₂ ([M+H]⁺): 280.2277, found: 280.2276.

3.5.2. (*S*)-*N*,*N*-*Diethyl*-2-((1*R*,*SS*)-5-*methyl*-7-oxo-4-*vinylcyclohept*-3-*enyl*)*propanamide* **58**. To a solution of **57** (12.3 mg, 44.0 µmol) in CH₂Cl₂ (0.44 mL) were added *N*-methylmorpholine oxide (NMO, 25.8 mg, 0.22 mmol) and molecular sieves 4 Å (22 mg) at room temperature. After 10 min at room temperature, tetrapropylammonium perruthenate (TPAP, 1.5 mg, 4.4 µmol) was added. After 30 min, the mixture was filtered and concentrated. The residue was purified by silica gel column chromatography (40% AcOEt/Hex) to give **58** (9.9 mg, 81%) as a colorless oil. [α]_D²³ +44.9 (*c* 0.29, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.06 (d, *J*=7.2 Hz, 3H), 1.10 (d, *J*=6.6 Hz, 3H), 1.13 (t, *J*=7.2 Hz, 3H), 1.20 (t, *J*=6.6 Hz, 3H), 2.06 (ddd, *J*=16.2, 12.0, 6.6 Hz, 1H), 2.28 (ddd, *J*=15.0, 9.6, 3.6 Hz, 1H), 2.63 (d, *J*=6.0 Hz, 2H), 2.84 (ddd, *J*=12.6, 10.2, 3.6 Hz, 1H), 3.00–3.07 (m, 2H), 3.32–3.45 (m, 4H), 4.96 (d, *J*=10.8 Hz, 1H), 5.13 (d, *J*=17.4 Hz, 1H), 5.83 (dd, *J*=9.0, 4.8 Hz, 1H), 6.23 (dd, *J*=17.4, 10.8 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ : 13.2 (q), 15.1 (q), 17.2 (d), 18.4 (t), 27.9 (t), 29.8 (d), 36.4 (d), 40.6 (t), 42.2 (t), 49.1 (t), 56.2 (d), 111.0 (t), 129.7 (d), 140.2 (d), 145.0 (q), 174.2 (q), 212.3 (q). IR (CHCl₃): 2975, 1701, 1627, 1463, 1220 cm⁻¹; MS (FAB) *m/z*: 278 ([M+H]⁺); HRMS (FAB) calcd for C₁₇H₂₈NO₂ ([M+H]⁺): 278.2120, found: 278.2117.

3.5.3. (S)-N,N-Diethyl-2-((1R,5S,7S)-7-hydroxy-5-methyl-4-vinylcyclohept-3-enyl)propanamide 59. To a solution of 58 (2.0 mg, 7.2 μmol) in THF (0.8 mL) were added Et₃N (5.0 μL, 36 μmol), H₂O (1 µL, 45 µmol), and SmI₂ (0.1 M in THF, 180 µL, 18.0 µmol) at 0 °C. The mixture was stirred for a few seconds at 0 °C, whereupon the reaction was quenched with saturated aqueous NH₄Cl, and then extracted with EtOAc. The combined organic layers were washed with saturated aqueous NaHCO3 and brine, and dried over MgSO4. After the mixture was evaporated, the residue was purified by silica gel column chromatography (50% AcOEt/Hex) to give 59 (1.6 mg, 80%) and the C8-isomer (20%) as colorless oils. Major isomer 59: $[\alpha]_{D}^{23}$ –80.0 (c 0.25, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.11–1.23 (m, 6H), 1.21 (t, J=7.2 Hz, 3H), 1.27 (d, J=7.2 Hz, 3H), 1.59-1.69 (m, 2H), 2.00-2.38 (m, 1H), 2.40 (ddd, J=16.2, 12.0, 4.8 Hz, 1H), 2.78 (ddd, J=14.4, 7.2, 2.4 Hz, 1H), 2.87–2.92 (m, 1H), 3.28–3.36 (m, 2H), 3.39-3.399 (m, 2H), 3.95-3.99 (m, 1H), 4.92 (d, J=10.8 Hz, 1H), 5.15 (d, *J*=17.4 Hz, 1H), 5.16 (s, 1H), 5.72 (dd, *J*=9.6, 4.8 Hz, 1H), 6.21 (dd, *I*=17.4, 10.8 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ: 12.7 (q), 13.0 (q), 14.9 (q), 17.1 (q), 28.6 (d), 31.1 (t), 40.9 (t), 41.5 (d), 42.0 (t), 42.5 (t), 46.5 (d), 69.5 (d), 110.2 (t), 129.3 (d), 140.1 (d), 146.7 (s), 177.1 (s). IR (CHCl₃): 2974, 2933, 1608, 1456 cm⁻¹; MS (FAB) m/z: 280 ([M+H]⁺); HRMS (FAB) calcd for C₁₇H₃₀NO₂ ([M+H]⁺): 280.2277, found: 280.2276.

3.5.4. (3S,3aR,7S,8aS)-3,7-Dimethyl-6-vinyl-3,3a,4,7,8,8a-hexahydro-2H-cyclohepta[b]furan-2-one **60**^{8c}. To a solution of **59** (16.5 mg, 59.1 µmol) in THF (0.47 mL) was added 3 M HCl (0.11 mL). The mixture was refluxed for 30 min. The reaction was quenched with saturated aqueous NaHCO₃, and extracted with Et₂O. The combined organic layers were washed with brine, and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (10% AcOEt/Hex) to give 60 (10.5 mg, 86%) as colorless prisms. Mp 82.5–83.5 °C; $[\alpha]_D^{23}$ –103.1 (*c* 0.32, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ: 1.13 (d, *J*=7.8 Hz, 3H), 1.22 (d, *J*=7.8 Hz, 3H), 1.70 (ddd, J=12.6, 12.6, 3.6 Hz, 1H), 2.07-2.16 (m, 2H), 2.28–2.34 (m, 2H), 2.69 (dq, J=7.8, 7.8 Hz, 1H), 3.07 (ddq, J=7.8, 4.2, 4.2 Hz, 1H), 4.53 (ddd, J=12.6, 10.2, 3.0 Hz, 1H), 4.96 (d, J=10.8 Hz, 1H), 5.17 (d, J=17.4 Hz, 1H), 5.81 (dd, J=9.6, 3.6 Hz, 1H), 6.23 (dd, J=17.4, 10.8 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ : 10.2 (q), 18.4 (q),25.0 (t), 28.1 (d), 36.5 (t), 40.0 (d), 46.2 (d), 81.4 (d), 110.3 (t), 129.8 (d), 141.4 (d), 145.4 (s), 180.0 (s). IR (KBr): 2958, 2929, 1766, 1627, 1465, 1450, 1211, 1039, 989, 887 cm⁻¹; MS (EI) *m/z*: 206 (M⁺), 79 (100%); HRMS (EI) calcd for C₁₃H₁₈O₂: 206.1307, found: 206.1302.

3.5.5. (–)-*Dihydroxanthatin* **2**⁸^{*c*}. To a solution of **60** (6.2 mg, 30 µmol) in dry CH₂Cl₂ (6.0 mL) were added the second generation Hoveyda–Grubbs catalyst **54** (1.9 mg, 3.0 µmol), and freshly distilled methyl vinyl ketone (49 µL, 0.60 mmol). The mixture was stirred at 45 °C for 1 h. Upon completion, the mixture was evaporated, and the residue was purified by column chromatography (30% EtOAc/Hex) to give dihydroxanthatin (**2**) (6.6 mg, 89%) as colorless prisms. Mp 121.0–124.0 °C (CH₂Cl₂/hexane); $[\alpha]_D^{23}$ –75.0 (*c* 0.32, CHCl₃); ¹H NMR (CDCl₃, 600 MHz) δ : 1.15 (d, *J*=7.8 Hz, 3H), 1.23 (d, *J*=7.8 Hz, 3H), 1.72 (ddd, *J*=12.6, 12.6, 3.6 Hz, 1H), 2.13 (dddd,

J=12.6, 12.6, 7.8, 2.4 Hz, 1H), 2.21 (ddd, *J*=12.0, 12.0, 3.0 Hz, 1H), 2.30 (s, 3H), 2.35 (ddd, *J*=12.6, 4.2, 4.2 Hz, 1H), 2.45 (ddd, *J*=16.2, 9.0, 2.4 Hz, 1H), 2.72 (dq, *J*=7.8, 7.8 Hz, 1H), 3.05 (ddq, *J*=7.8, 4.2, 4.2 Hz, 1H), 4.54 (ddd, *J*=12.6, 10.8, 3.0 Hz, 1H), 6.18 (d, *J*=16.2 Hz, 1H), 6.25 (dd, *J*=9.6, 3.0 Hz, 1H), 7.05 (d, *J*=16.2 Hz, 1H); ¹³C NMR (CDCl₃, 150 MHz) δ : 10.2 (q), 18.4 (q), 25.7 (t), 27.9 (q), 29.0 (d), 36.4 (t), 39.9 (d), 45.9 (d), 80.8 (d), 124.4 (d), 139.3 (d), 144.5 (s), 148.4 (d), 179.0 (s), 198.6 (s). IR (KBr): 2968, 1768, 1679, 1585, 1357, 1282, 1205, 1180, 981 cm⁻¹; MS (EI) *m/z*: 248 (M⁺), 248 (100%); HRMS (EI) calcd for C₁₅H₂₀O₃: 248.1412, found: 248.1417.

3.6. Synthesis of (-)-xanthatin (3)

3.6.1. N,N-Diethyl-2-((1R,5S,7R)-7-hydroxy-5-methyl-4-vinylcyclohept-3-enyl)acrylamide 63. To a solution of 53 (6.4 mg, 31.3 μ mol) in CH₂Cl₂ (0.5 mL) were added AlCl₃ (8.4 mg, 62.7 μ mol) and Et₂NH (6.5 µL, 62.7 µmol) at 0 °C. The mixture was stirred for 30 min at room temperature, whereupon the reaction was quenched with 1 M HCl, and the mixture was extracted with CH₂Cl₂. The combined organic layers were washed with saturated aqueous NaHCO₃ and brine, and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (50% AcOEt/Hex) to give 63 (8.4 mg, 97%) as a colorless oil. $[\alpha]_D^{20}$ –58.3 (c 0.06, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.18 (t, *J*=7.2 Hz, 6H), 1.25 (d, *J*=7.2 Hz, 3H), 1.78 (ddd, *J*=14.4, 4.8, 2.4 Hz, 1H), 1.86 (ddd, *J*=14.4, 9.6, 2.4 Hz, 1H), 2.07 (ddd, *J*=14.4, 7.2, 7.2 Hz, 1H), 2.68 (br d, J=12.0 Hz, 6H), 2.78 (ddd, J=14.4, 14.4, 4.2 Hz, 1H), 2.83 (dq, J=7.2, 7.2 Hz, 1H), 3.339 (dq, J=14.4, 7.2 Hz, 1H), 3.343 (dq, J=14.4, 7.2 Hz, 1H), 3.55 (dq, J=14.4, 7.2 Hz, 2H), 4.15–4.17 (m, 1H). 4.91 (d, *J*=10.8 Hz, 1H), 4.96 (br s, 1H), 5.126 (d, *J*=16.8 Hz, 1H), 5.132 (s, 1H), 5.26 (s, m), 5.73 (dd, J=9.6, 4.8 Hz, 1H), 6.21 (dd, J=16.8, 10.8 Hz, 1H); 13 C NMR (150 MHz, CDCl₃) δ : 12.6 (q), 14.2 (q), 20.3 (q), 25.0 (t), 31.4 (d), 38.4 (t), 39.0 (t), 43.3 (t), 48.5 (d), 74.4 (d), 110.4 (t), 115.8 (t), 129.0 (d), 140.4 (d), 146.5 (s), 146.7 (s), 172.2 (s). IR (CHCl₃): 2927, 1596, 1458 cm⁻¹; MS (FAB) *m*/*z*: 278 ([M+H]⁺); HRMS (FAB) calcd for C₁₇H₂₈NO₂ ([M+H]⁺): 278.2120, found: 278.2120.

3.6.2. N,N-Diethyl-2-((1R,5S)-5-methyl-7-oxo-4-vinylcyclohept-3enyl)acrylamide 64. To a solution of 63 (7.1 mg, 25.6 µmol) in CH₂Cl₂ (0.3 mL) were added NMO (15.0 mg, 128.0 µmol) and molecular sieves 4 Å (12.8 mg) at room temperature. The mixture was stirred for 10 min, and then TPAP (0.9 mg, 2.6 µmol) was added. The mixture was stirred for 30 min. After filtration, the filtrate was purified by silica gel column chromatography (40% AcOEt/Hex) to give **64** as a colorless oil. $[\alpha]_D^{20}$ +11.6 (*c* 0.17, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ: 1.15–1.17 (m, 9H), 2.43 (ddd, *J*=15.6, 9.6, 4.2 Hz, 1H), 2.61 (dd, J=10.8, 5.4 Hz, 1H), 2.75-2.80 (m, 1H), 2.92-3.01 (m, 2H), 3.4 (br s, 3H), 3.47 (dd, J=13.2, 4.2, 4.2 Hz, 1H), 3.52 (br s, 1H), 4.98 (d, J=10.8 Hz, 1H), 5.13 (d, J=17.4 Hz, 1H), 5.22 (s, 1H), 5.26 (d, *J*=1.2 Hz, 1H), 5.86 (dd, *J*=9.6, 4.8 Hz, 1H), 6.20 (dd, *J*=17.4, 10.8 Hz, 1H); 13 C NMR (150 MHz, CDCl₃) δ : 12.6 (q), 14.2 (q), 20.0 (q), 27.6 (t), 31.2 (d), 38.8 (d), 43.0 (d), 47.3 (t), 57.1 (d), 111.8 (t), 115.3 (t), 128.4 (d), 139.7 (d), 142.9 (s), 144.6 (s), 170.3 (s), 210.3 (s). IR (CHCl₃): 2999, 1703, 1610, 1461 cm⁻¹; MS (FAB) *m/z*: 276 ([M+H]⁺); HRMS (FAB) calcd for C₁₇H₂₆NO₂ ([M+H]⁺): 276.1964, found: 276.1966.

3.6.3. (3aR,75,8aS)-7-Methyl-3-methylene-6-vinyl-3,3a,4,7,8,8ahexahydro-2H-cyclohepta[b] furan-2-one **67**. To a solution of **64** (4.4 mg, 16.0 µmol) in THF (0.3 mL) at 0 °C was added DIBAL-H (18.9 µL, 17.6 µmol). The resulting mixture was stirred for 0.5 h at room temperature, then cooled to 0 °C, whereupon water was added. The mixture was filtered and the filtrate was dried over MgSO₄. The crude product (a mixture of the diastereomers, **65:63**=1:1.2) was used directly in the next step. *Compound* **65**: $[\alpha]_D^{20}$ –141.7 (*c* 0.06, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.12 (d, *J*=7.8 Hz, 3H), 1.18 (t, *J*=6.6 Hz, 6H), 1.59–1.61 (m, 1H), 2.14 (ddd, *J*=13.2, 4.8, 2.4 Hz, 1H), 2.17–2.22 (m, 2H), 2.27–2.33 (m, 1H), 2.93–2.96 (m, 1H), 3.40 (dd, *J*=15.6 Hz, 1H), 2.61 (dd, *J*=10.8, 5.4 Hz, 1H), 2.75–2.80 (m, 1H), 2.92–3.01 (m, 2H), 3.40 (dq, *J*=14.4, 7.2 Hz, 2H), 3.49–3.57 (m, 2H), 3.9 (br t, *J*=10.8 Hz, 1H), 4.64 (d, *J*=3.0 Hz, 1H), 4.95 (d, *J*=10.8 Hz, 1H), 5.15 (s, 1H), 5.17 (d, *J*=17.4 Hz, 1H), 5.29 (s, 1H), 5.71 (dd, *J*=9.6, 4.2 Hz, 1H), 6.22 (dd, *J*=17.4, 10.8 Hz, 1H).

To a solution of the diastereomeric mixture (65 and 63) $(16.0 \,\mu mol)$ in THF $(320 \,\mu L)$ was added 3 M HCl $(80 \,\mu L)$. The mixture was refluxed for 30 min. The reaction was guenched with saturated aqueous NaHCO₃ and extracted with Et₂O. The combined organic layers were washed with brine and dried over MgSO₄. After the mixture was evaporated, the residue was purified by silica gel column chromatography (10% AcOEt/Hex) to give 67 (1.3 mg, 27% for two steps) and **53** (33% for two steps) as colorless needles. *Compound* **67**: mp 77.2–77.7 °C (hexane); [α]_D²³ –40.0 (*c* 0.05, CHCl₃); ¹H NMR (600 MHz, CDCl₃) δ : 1.14 (d, *J*=7.8 Hz, 3H), 1.83 (ddd, J=12.6, 12.6, 3.6 Hz, 1H), 2.14 (ddd, J=16.2, 11.4, 3.6 Hz, 1H), 2.35 (ddd, J=13.2, 4.2, 3.6 Hz, 1H), 2.51-2.56 (m, 1H), 2.67 (ddd, J=16.2, 9.0, 2.4 Hz, 2H), 3.11 (dq, J=3.6, 3.6 Hz, 1H), 4.29 (ddq, *J*=12.6, 10.2, 2.4 Hz, 1H), 4.99 (d, *J*=11.4 Hz, 1H), 5.20 (d, *J*=17.4 Hz, 1H), 5.47 (dd, J=9.0, 3.0 Hz, 1H), 6.18 (d, J=3.0 Hz, 1H), 6.26 (dd, J=17.4, 10.8 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ: 18.9 (q), 26.6 (t), 28.3 (d), 36.7 (t), 47.9 (d), 82.1 (d), 110.7 (t), 118.6 (t), 128.9 (d), 139.7 (s), 141.5 (d), 145.7 (s), 170.1 (s). IR (CHCl₃): 3005, 1764, 1600 cm⁻¹; MS (EI) *m*/*z*: 204 (M⁺), 204 (100%); HRMS (EI) calcd for C₁₃H₁₆O₂: 204.1150, found: 204.1151.

3.6.4. (–)-Xanthatin 3^{29} . To a solution of **67** (3.0 mg, 14.7 µmol) in dry CH₂Cl₂ (3.0 mL) were added the second generation Hoveyda-Grubbs catalyst 54 (0.9 mg, 1.5 µmol), and freshly distilled methyl vinyl ketone (11.9 µL, 147 µmol). The mixture was stirred at 45 °C for 1 h. Upon completion, the mixture was evaporated, and the residue was purified by column chromatography (30% EtOAc/Hex) to give xanthatin (3) (3.0 mg, 83%) as colorless needles. Mp 112.1–113.1 °C (Et₂O/hexane); lit.^{8d} mp 114.5–115.2 °C (EtOH); $[\alpha]_D^{25} - 20.0 (c \ 0.15, CHCl_3)$; $[it.^{8d} [\alpha]_D - 17.8 (c \ 0.14, CHCl_3)$; ¹H NMR (600 MHz, CDCl₃) δ : 1.17 (d, J=7.8 Hz, 3H), 186 (ddd, J=12.6, 12.6, 3.6 Hz, 1H), 2.22 (ddd, J=12.6, 9.6, 3.6 Hz, 1H), 2.31 (s, 3H), 2.38 (ddd, J=12.6, 4.2, 2.4 Hz, 1H), 2.53-2.58 (m, 1H), 2.80 (ddd, J=16.2, 8.4, 2.4 Hz, 1H), 2.80-2.86 (m, 1H), 3.39-3.44 (m, 1H), 4.65 (ddd, J=12.0, 8.4, 2.4 Hz, 1H), 3.09 (ddq, J=8.4, 4.2, 4.2 Hz, 1H), 4.30 (ddd, J=12.6, 9.6, 2.4 Hz, 1H), 5.49 (d, J=3.0 Hz, 1H), 6.207 (d, J=16.2 Hz, 1H), 6.212 (d, J=3.0 Hz, 1H), 6.29 (dd, J=9.6, 2.4 Hz, 1H), 7.08 (d, J=16.2 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ: 18.9, 27.2, 27.9, 29.2, 36.6, 47.5, 81.5, 119.0, 124.7, 138.1, 139.2, 144.8, 148.5, 169.7, 198.5. IR (KBr): 2927, 1760, 1685, 1587, 1402, 1353, 1247, 1178, 1137, 977 cm⁻¹; MS (EI) *m*/*z*: 246 (M⁺), 91 (100%); HRMS (EI) calcd for C₁₅H₁₈O₃ (M⁺): 246.1256, found: 246.1258.

Acknowledgements

This research was partially supported by a Grant-in-Aid for Scientific Research on (B) (19390007, 22390002) and the Program for Promotion of Basic and Applied Research for Innovations in the Biooriented Industry (BRAIN). K.M. additionally acknowledges the support of the JSPS. We are also grateful to Ms. C. Moriya for experimental contribution in the early stage of this project.

Supplementary data

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.tet.2010.08.061.

References and notes

- 1. (a) Ohno, S.; Tomita-Yokotani, K.; Kosemura, S.; Node, M.; Suzuki, T.; Amano, M.; Yasui, K.; Goto, T.; Yamamura, S.; Hasegawa, K. Phytochemistry 2001, 56, 577-581: (b) Tomita-Yokotani, K.: Kato, I.: Kosemura, S.: Yamamura, S.: Kushima, M.; Kakuta, H.; Hasegawa, K. *Phytochemistry* **1997**, *46*, 503–506; (c) Yo-kotani-Domitian, K.; Kato, J.; Yamada, K.; Kosemura, S.; Yamamura, S.; Bruinsma, J.; Hasegawa, K. *Physiol. Plant.* **1999**, *106*, 326–330.
- (a) Ahn, J.-W.; No, Z.; Ryu, S.-Y.; Zee, O.-P.; Kim, S.-K. Nat. Prod. Sci. 1995, 1, 1-4; (b) Kovács, A.; Vasas, A.; Forgo, P.; Réthy, B.; Zupkó, I.; Hohmann, J. Z. Naturforsch., C: J. Biosci. 2009, 64c, 343-349; (c) Ramírez-Erosa, I.; Huang, Y.; Hickie, R. A.; Sutherland, R. G.; Barl, B. Can. J. Physiol. Pharmacol. 2007, 85, 1160-1172; (d) Kim, Y. S.; Kim, J. S.; Park, S.-H.; Choi, S.-U.; Lee, C. O.; Kim, S.-K.; Kim, Y.-K.; Kim, S. H.; Ryu, S. Y. Planta Med. 2003, 69, 375-377.
- 3 Joshi, S. P.; Rojatkar, S. R.; Nagasampagi, B. A. J. Med. Aromat. Plant Sci. 1997, 19, 366-368
- 4 (a) Kato, T.; Yokotani-Tomita, K.; Suzuki, T.; Kosemura, S.; Hasegawa, K. Weed Biol. Manag. 2008, 3, 124-128; (b) Lavault, M.; Landreau, A.; Larcher, G.; Bouchara, J.-P.; Pagniez, F.; Pape, P. L.; Richomme, P. Fitoterapia 2005, 76, 363-366; (c) Ginesta-Peris, E.; Garcia-Breijo, F. J.; Primo-Yúfera, E. Lett. Appl. Microbiol. 1994, 18, 206-208.
- (a) Little, J. E.; Foote, M. W.; Johnstone, D. B. Arch. Biochem. 1950, 27, 247-254; 5 (b) Geissman, T. A.; Deuel, P.; Bonde, E. K.; Addicott, F. A. J. Am. Chem. Soc. 1954, 76, 685-687; (c) Deuel, P.; Geissman, T. A. J. Am. Chem. Soc. 1957, 79, 3778-3783.
- 6. Sato, Y.; Oketani, H.; Yamada, T.; Shingyouchi, K.; Ohtsubo, T.; Kihara, M.; Shibata, H.; Higuti, T. J. Pharm. Pharmacol. 1997, 49, 1042-1044.
- 7. (a) McMillan, C.; Chavez, P. I.; Mabry, T. J. Biochem. Syst. Ecol. 1975, 3, 137-141; (b) Kawazu, K.; Nakajima, S.; Ariwa, M. Experientia 1979, 35, 1294-1295; (c) Bohlmann, F.; Singh, P.; Joshi, K. C.; Singh, C. L. Phytochemistry 1982, 21, 1441-1443; (d) Ghazy, N. M.; Omar, A. A.; Elrashidy, E. M.; Metwally, A. M. Egypt. J. Pharm. Sci. 1988, 29, 39-42; (e) Ahmed, A. A.; Jakupovic, J.; Bohlmann, F.; Regaila, H. A.; Ahmed, A. M. Phytochemistry 1990, 29, 2211-2215.
- (a) For total syntheses of (+)-8-*epi*-xanthatin, see Kummer, D. A.; Brenneman, J. B.; Martin, S. F. *Org. Lett.* **2005**, *7*, 4621–4623; (b) Kummer, D. A.; Brenneman, J. B.; Martin, S. F. Tetrahedron 2006, 62, 11437-11449 for (-)-dihydroxanthatin, see: (c) Evans, M. A.; Morken, J. P. Org. Lett. 2005, 7, 3371-3373 for (-)-xanthatin, see: (d) Yokoe, H.; Yoshida, M.; Shishido, K. Tetrahedron Lett. 2008, 49, 3504-3506 for a review, see: (e) Shishido, K. Heterocycles 2009, 78, 873-889.
- 9. For total syntheses of (+)-sundiversifolide, see: (a) Yokoe, H.; Sasaki, H.; Yoshimura, T.; Shindo, M.; Yoshida, M.; Shishido, K. Org. Lett. 2007, 9, 969-971; (b) Ohtsuki, K.; Matsuo, K.; Yoshikawa, T.; Moriya, C.; Tomita-Yokotani, K.; Shishido, K.; Shindo, M. Org. Lett. 2008, 10, 1247-1250 (±)-sundiversifolide,

see: (c) Hashimoto, T.; Tashiro, T.; Sasaki, M.; Takikawa, H. Biosci. Biotechnol. Biochem. 2007, 71, 2046–2051 (-)-diversifolide, see: (d) Matsuo, K.; Yokoe, H.; Shishido, K.; Shindo, M. Tetrahedron Lett. 2008, 49, 4279-4281.

- 10. For a review, see: Molander, G. A. Acc. Chem. Res. 1998, 31, 603-609
- I. Evans, D. A.; Ennis, M. D.; Mathre, D. J. J. M. Chem. Soc. 1982, 104, 1737–1739,
 Sharpless, K. B.; Amberg, W.; Bennani, Y. L.; Crispino, G. A.; Hartung, J.; Jeong, K.-S.; Kwong, H.-L.; Morikawa, K.; Wang, Z.-M.; Zhang, X.-L. J. Org. Chem. 1992, 57, 2768-2771.
- For cyclization of iodo esters with organolithiums, see: (a) Cooke, M. P., Jr.; 13 Houpis, I. N. Tetrahedron Lett. **1985**, 26, 4987–4990; (b) Saito, T.; Takeuchi, T.; Matsuhashi, M.: Nakata, T. Heterocycles 2007, 72, 151-156.
- 14. The conformational analysis was performed with MMFF force field (CONFLEX v. 6)
- 15. Higashibayashi, S.; Shinko, K.; Ishizu, T.; Hashimoto, K.; Shirahama, H.; Nakata, M. Synlett 2000, 1306-1308.
- Without use of CH₂Cl₂ as the co-solvent, the diol was mainly generated. See 16. Ref 15
- 17. The keto group on **36** was inert to olefinations, such as the Wittig or the Peterson reactions, probably due to the steric hindrance of the TBS and TBDPS groups.
- Garner, P.; Ramakanth, S. J. Org. Chem. 1987, 52, 2629-2631. 18
- Kido, F.; Tsutsumi, K.; Maruta, R.; Yoshikoshi, A. J. Am. Chem. Soc. 1979, 101, 19. 6420-6424.
- Finkielsztein, L. M.; Aguirre, J. M.; Lantano, B.; Alesso, E. N.; Moltrasio Iglesias, 20 G. Y. Synth. Commun. 2004, 34, 895-901.
- Bates, R. W.; Fernandez-Megia, E.; Ley, S. V.; Ruck-Braun, K.; Tilbrook, D. M. G. J. 21
- Chem. Soc., Perkin Trans. 1 1999, 1917-1925. 22
- Suzuki, H.; Fuchita, T.; Iwasa, A.; Mishina, T. Synthesis 1978, 905-908. Behan, J. M.; Johnstone, R. A. W.; Wright, M. J. J. Chem. Soc., Perkin Trans. 1 1975, 23. 1216-1217
- Higuchi, Y.; Shimoma, F.; Ando, M. J. Nat. Prod. 2003, 66, 810-817. 24
- 25. Garber, S. B.; Kingsbury, J. S.; Gray, B. L.; Hoveyda, A. H. J. Am. Chem. Soc. 2000, 122 8168-8179
- 26. Shishido successfully carried out the Mitsunobu reaction at C-8 by using a less sterically hindered substrate. See Ref. 8d.
- 27 Dahlen, A.; Hilmersson, G. Chem.-Eur. J. 2003, 9, 1123-1128.
- The stereochemistry was speculated based on the Shishido's stereochemical 28. result in the reaction of diphenyl diselenide and lithium enolate of 60. See Ref. **b**8
- 29. (a) Marco, J. A.; Sanz-Cervera, J. F.; Corral, J.; Carda, M.; Jakupovic, J. Phytochemistry 1993, 34, 1569-1576; (b) Pinel, B.; Audob, G.; Mallet, S.; Lavault, M.; De La Poype, F.; Séraphin, D.; Richommea, P. J. Chromatogr., A 2007, 1151, 14-19.
- 30. Nacro, K.; Baltas, M.; Gorrichon, L. Tetrahedron 1999, 55, 14013-14030.